From Verification to Guided-Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?

Pamela Sue Scott

Grand Valley State University

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From Verification to Guided-Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?

Pamela Sue Scott

A Thesis Submitted to the Graduate Faculty of
GRAND VALLEY STATE UNIVERSITY

In
Partial Fulfillment of the Requirements
For the Degree of
Master of Education

Advanced Content Specialization - Chemistry

August 2014
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Pamela S. Scott
Abstract

How does the degree or level of inquiry-based laboratory instruction impact student performance and student perseverance in the laboratory portion of a first-semester general chemistry course? In 2008, a two-year community college sought to answer this question by replacing the traditional verification laboratory curriculum with a guided-inquiry laboratory curriculum. This change provided a case study of the 'new' guided-inquiry curriculum vs. the 'old' traditional verification curriculum. Researchers used a modified for college instruction version of The Continuum of Scientific Inquiry Rubric (Fay, Grove, Towns, & Bretz, 2007) to assess both laboratory curricula, to determine the level of inquiry incorporated into each laboratory experiment as well as the inquiry levels of both laboratory curricula overall. Student performance was evaluated via laboratory report average final grades and individual laboratory report scores, while student perseverance was measured by comparing overall completion rates of laboratory reports and student withdrawal rates for each laboratory curriculum to determine if any relationships exist between level(s) of inquiry and student performance and student perseverance.
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Chapter One: Introduction

Problem Statement

Increasing the use of inquiry-based science instructional activities in both lecture and laboratory classrooms has been at the heart of most calls for curriculum reform at all levels of education (Bransford, Brown & Cocking, 2000; Lederman, 2004). While many studies have been conducted on the effectiveness of inquiry-based science instruction techniques with respect to student performance and outcomes in the chemistry laboratory, literature reveals that the definition of inquiry-based science instruction lacks consistency, and a common standardized performance measure for assessing the value of inquiry-based laboratory instruction remains variable. Furthermore, it is difficult to compare studies that investigate the effectiveness of inquiry-based teaching in the laboratory because the degree or level of inquiry present is most often overlooked and seldom characterized or reported.

Importance and Rationale

Science education is in the midst of a crucial transition, as an inquiry-based approach to instruction improves student achievement (Blanchard et al., 2010) and allows students to make connections and correlations between science and their own lives, which can be particularly important for culturally diverse learners (Adamson et al., 2003). Inquiry-based laboratory instruction promotes the development of students’ scientific processing skills and can advance critical thinking capabilities, problem-solving skills, and an overall understanding of the nature of science (Hofstein & Lunetta, 2004). The National Science Teachers Association asserts “understanding science content is significantly enhanced when ideas are anchored to inquiry experiences.” Therefore, “scientific inquiry is an effective and powerful way of understanding science content” (National Science Teachers Association, 2004).
The National Science Education Standards (National Research Council, 2000) and the Benchmarks for Science Literacy (American Association for the Advancement of Science, 2009) emphasize the importance of inquiry-based instruction, where classrooms are student-centered, and laboratory activities are hands-on and minds-on, allowing students to engage in the processes that scientists use in constructing knowledge. Inquiry-based science instruction should not be an isolated occurrence, but a comprehensive and ongoing pedagogical approach.

**Background**

In 1986, the National Science Board (NSB) reported that laboratory instruction “has deteriorated to the point where it is often uninspired, tedious and dull,” and courses and curricula “fail to reflect advances in the understanding of teaching and learning” (National Science Board, 1986, p. 2). Even so, the predominant type of laboratory instruction in today’s chemistry curriculum is still the traditional verification style, also known as cookbook chemistry. Within the verification laboratory environment the teacher presents a study topic, the students follow a structured procedure, or recipe, in order to collect data and the final outcome is predetermined known results. The students are involved in neither the planning of the scientific investigation nor the interpretation of the results; hence the traditional method of laboratory instruction has limited opportunities for students to understand science content. Furthermore, most textbooks are filled with information and terminology that students are expected to memorize, and most exams assess students’ abilities to recall the facts. This type of instruction places very little emphasis on critical thinking and is an unrealistic representation of scientific experimentation (Domin, 1999).
The Inquiry Process

Inquiry-based instruction is not a new technique, but it does stand in stark contrast to the more structured, traditional lecture-and-test instruction and verification centered laboratory curriculum framework of today's schools. Brunner (2012) of the Center for Children and Technology classifies the inquiry process into four stages, as illustrated in Figure 1.0.

Figure 1.0 The Inquiry Process.

![Diagram of the Inquiry Process]

Stage (1) Formulation (pose real questions). Inquiry-based instruction requires a classroom teacher to play a much different role than that of a teacher in a traditional classroom. Instead of providing direct chalk-and-talk instruction to students, teachers help students create
their own content related questions. Thinking is not driven by answers but by questions; thus, teaching students to ask the right questions is one of the greatest skills a teacher can impart.

**Stage (2) Exploration (find resources).** Teachers facilitate students as they develop skills and learn how to filter the vast resources (i.e. internet, books, journals, people, media) to find the information they need.

**Stage (3) Collection (interpret information).** Teachers assist students in evaluating the resources for accuracy and validity, and in processing the information to form conclusions.

**Stage (4) Assessment (report findings).** Teachers support and encourage students as they learn to create meaningful representations of their research findings and transfer information skills and knowledge to solve problems and make decisions (Brunner, 2012).

Likewise, the National Science Education Standards describe inquiry-based science instruction as involving students in active learning that emphasizes questioning, data analysis, and critical thinking.

Students at all grade levels and in every domain of science should have the opportunity to use scientific inquiry and develop the ability to think and act in ways associated with inquiry, including asking questions, planning and conducting investigations, using appropriate tools and techniques to gather data, thinking critically and logically about relationships between evidence and explanations, constructing and analyzing alternative explanations, and communicating scientific arguments. (National Research Council, 1996, p. 105)

Although a considerable body of literature supports inquiry-based science instruction, little progress has been made to integrate inquiry-based teaching in the science laboratory, particularly at the college and university levels (Bruck, Bretz & Towns, 2008). Today, the vast
majority of higher education institutions continue to use a more structured, traditional verification laboratory curriculum as the primary means of educating students in the science laboratory (Abraham et al., 1997). College faculty often resist inquiry-based instruction because they view inquiry as costly, time consuming, student-directed, and as such, potentially chaotic. In addition, they often perceive significant obstacles (i.e. class sizes, limited resources, student competencies) when incorporating inquiry into laboratory activities (Brown, Abell, Demir & Schmidt, 2006).

Analysis of the literature indicates that the effectiveness of inquiry-based laboratory instruction has been measured by many different performance measures, such as interviews, observations, surveys, and questionnaires which are designed to capture changes in student attitudes, beliefs and perceptions with respect to the laboratory environment. In addition, few studies report quantitative comparisons of exam scores that have been used to evaluate student achievement in the laboratory.

Further complicating the assortment of performance measures and student outcomes is the fact that it is difficult to compare studies that investigate the effectiveness of inquiry-based instruction, because the degree or level of inquiry present in a particular activity, laboratory experiment or curriculum is most often not characterized. As if in answer to this limitation, a Continuum of Scientific Inquiry Rubric has been developed. The rubric was first described by Schwab (1960) as “three levels of openness and permissiveness,” and later amended by Herron (1971) into “Herron’s scale: a four-point scale describing four levels of inquiry.” Subsequently, the rubric was adapted and used by Lederman (2004) for high school instruction, and more recently, the rubric has been modified for college instruction (Fay, Grove, Towns, & Bretz, 2007). Rather than seeking to answer the question of whether inquiry-based instruction had
value in laboratory application, Fay et al. (2007) focused on showing that the inquiry content of a laboratory curriculum could be quantitatively measured. Their study confirms that a modified, for college instruction, rubric (Table 1.0) and rubric companion (Table 1.1), depicting a visual comparison of inquiry characteristics, can be consistently used to evaluate the level of inquiry in laboratory activities by multiple different raters, and thus can aid instructors in making laboratory curricula choices. In addition, an article by Bruck et al. (2008) suggests that the rubric can be used to standardize the means of communication with respect to inquiry-based science instruction and learning in future higher education research literature.

**Table 1.0 The Continuum of Scientific Inquiry Rubric** (Fay et al., 2007).

<table>
<thead>
<tr>
<th>Level of Inquiry</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>Level-0</td>
<td>The problem, procedure and methods to solutions are provided to the student. The student performs the experiment and verifies the results with the manual.</td>
</tr>
<tr>
<td>Level-1</td>
<td>The problem and procedure are provided to the student. The student interprets the data in order to propose viable solutions.</td>
</tr>
<tr>
<td>Level-2</td>
<td>The problem is provided to the student. The student develops a procedure for investigating the problem, decides what data to gather, and interprets the data in order to propose viable solutions.</td>
</tr>
<tr>
<td>Level-3</td>
<td>A ’raw’ phenomenon is provided to the student. The student chooses the problem to explore, develops a procedure for investigating the problem, decides what data to gather, and interprets the data in order to propose viable solutions.</td>
</tr>
</tbody>
</table>

**Table 1.1 Rubric Companion: Visual Comparison of Inquiry Characteristics**

<table>
<thead>
<tr>
<th>Level</th>
<th>Problem/Question</th>
<th>Procedure/Method</th>
<th>Solution</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>Provided to student</td>
<td>Provided to student</td>
<td>Provided to student</td>
</tr>
<tr>
<td>1</td>
<td>Provided to student</td>
<td>Provided to student</td>
<td>Constructed by student</td>
</tr>
<tr>
<td>2</td>
<td>Provided to student</td>
<td>Constructed by student</td>
<td>Constructed by student</td>
</tr>
<tr>
<td>3</td>
<td>Constructed by student</td>
<td>Constructed by student</td>
<td>Constructed by student</td>
</tr>
</tbody>
</table>

**Grand Rapids Community College**

Grand Rapids Community College (GRCC) always used a traditional laboratory curriculum in the general chemistry two-semester sequence (CM103/104), which consisted of a
collection of verification experiments written in house by the chemistry faculty. However, in 2008 the chemistry department actively engaged in a laboratory curriculum evaluation of the two-semester general chemistry course. In an effort to provide a more dynamic learning environment, the faculty decided to implement a commercially published guided-inquiry curriculum. The inquiry-based laboratory curriculum that is the focus of this research is *Guided Inquiry Experiments for General Chemistry: Practical Problems and Applications* (Kerner & Lamba, 2008).

**Statement of Purpose**

The research presented in this thesis makes use of Fay et al.'s (2007) modified for college instruction rubric (Table 1.0) to characterize the level of inquiry present in both the traditional and inquiry-based laboratory curricula, to examine the relationship between the level of inquiry present in a laboratory experience and student outcomes, and to use a standard means of communication with respect to inquiry-based science instruction. In addition, this investigation documents the changes observed when a laboratory program is transitioned from a structured traditional laboratory curriculum to a more inquiry-based instructional laboratory curriculum, thus informing and providing quantitative data and evidence to chemistry instructors, science educators, schools and institutions as they contemplate inquiry-based science curriculum reform.

**Research Question**

How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at Grand Rapids Community College (GRCC)? In 2008, a commercially published guided-inquiry laboratory curriculum (Kerner & Lamba, 2008) was implemented for the first-semester general chemistry course (CM103) at Grand Rapids Community College in Michigan.
This curriculum change provided a case study of the new guided-inquiry laboratory curriculum versus the old traditional laboratory curriculum.

**Hypothesis**

The implementation of a more inquiry-based instructional laboratory curriculum will positively impact student performance and student perseverance in the laboratory portion of the first-semester general chemistry course (CM103) designed to educate students majoring in science and engineering programs.

**Design, Data Collection, and Analysis**

A quasi-experimental design and nonrandom sampling procedure was used to establish two student test groups from intact laboratory sections. Comparison of ACT test scores was used to ascertain that the two samples of students were drawn from the same population. The student performance data set consisted of students’ individual laboratory report scores and overall laboratory report final grades, three semesters from the old traditional curriculum (Fall 2006, Winter 2007, and Fall 2007) and three semesters from the new guided-inquiry curriculum (Fall 2008, Winter 2009, and Fall 2009). Additionally, a comparison of laboratory report completion rates between the two laboratory curricula provided a measure of student perseverance in the laboratory portion of the course. In an effort to minimize instructor effect, student data was provided by the same laboratory instructor for all six semesters.

The level of inquiry present in each experiment for both laboratory curricula was determined using the modified version of The Continuum of Scientific Inquiry Rubric (Fay et al., 2007) for college instruction (Table 1.0).

Statistical analysis was done using nonparametric versions of the t-test and Analysis of Variance procedures to determine the difference in levels of inquiry present in each laboratory.
curriculum, effect of level of inquiry on student performance, a comparison of laboratory report completions rates and a comparison of withdrawal grades as a measure of student perseverance in the laboratory portion of the course. A Kolmogorov-Smirnov test (Field, 2005) was used to check for normality of data and determined that nonparametric statistical techniques were necessary because these data did not exhibit normal distribution.

Definitions

Inquiry-based science instruction, also known as teaching science through inquiry, active learning, discovery learning, or simply as scientific inquiry, combines the curiosity of students with the application of scientific processes to enhance the development of critical thinking skills (Figure 1.1). As students encounter problems they do not understand, they develop questions, make observations, collect and interpret data, and apply new information to propose possible solutions (National Research Council, 2000).

Figure 1.1 Inquiry-Based Learning Cycle (Carin, Bass & Contant, 2004, p. 21).
Inquiry-based science instruction is a pedagogical approach that gives students opportunities to take ownership of their own learning, a skill necessary to succeed in society. Moreover, scientific inquiry is the process used by scientists to investigate the nature of science, and it is at the core of how students learn or create knowledge. The National Science Education Standards define scientific inquiry as,

The diverse ways in which scientists study the natural world and propose explanations based on the evidence derived from their work. Inquiry also refers to the activities of students in which they develop knowledge and understanding of scientific ideas, as well as an understanding of how scientists study the natural world. (National Research Council, 1996, p. 23)

Delimitations of the Study

Student performance was evaluated via laboratory report average final grades and individual laboratory report scores. A comparison of laboratory report completion rates between the two laboratory curricula provided a measure of student perseverance. Thus, in an effort to minimize instructor effect, laboratory report data was accumulated and provided by the same laboratory instructor throughout the data collection segment of this study.

Limitations of the Study

This research was unable to correlate the changes in the laboratory format to improvements in student learning in chemistry because of institutional course grading limitations. Without external, course objective-related measures of learning, claims about the impact on student learning due to the implementation of the inquiry-based laboratory curriculum could not be rendered. Therefore, this research used and compared students’ individual
laboratory report scores and laboratory report average final grades as a measure of student performance in the two laboratory curricula.

**Organization of the Thesis**

Chapter Two provides an interpretation of the theoretical framework and presents an analysis of current research concerned with inquiry-based laboratory instruction. Chapter Three focuses on the key features (participants, instrumentation, data collection and analysis) with respect to this study’s research method. Chapter Four reports the findings and interprets the student performance and student perseverance results in terms of the research question, “How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at GRCC?” and hypothesis, “The implementation of a more inquiry-based instructional laboratory curriculum will positively impact student performance and student perseverance in the laboratory portion of the first-semester general chemistry course (CM103) designed to educate students majoring in science and engineering programs.” Chapter Five brings this study to a close with an explanation, recommendation, and articulation of the results.
Chapter Two: Literature Review

Introduction

This chapter considers the research question, “How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at GRCC?” and provides a synthesis of current research with respect to inquiry-based laboratory instruction. Thus, Chapter Two comprehensively explores a variety of studies that use a number of different descriptions with respect to inquiry-based science instruction and reviews an assortment of performance measures used by researchers to assess student outcomes and the value of inquiry-based science instruction in the chemistry laboratory. As a result, this chapter (1) provides a description of the theoretical framework, (2) presents several peer-reviewed research studies, (3) summarizes the findings, and (4) closes with a brief conclusion.

Theoretical Framework

The postpositivist philosophy is consistent with the approach of traditional instruction and verification laboratory curriculum, in which it is assumed that as long as the classroom material is presented; the students will learn. In this case, the instruction is the cause and the learning is the effect. The students have a responsibility to learn, but it is the presentation of information by the teacher that leads to the learning (Creswell, 2009).

By contrast, the social constructivist perspective holds that individuals construct meaning via interacting with each other and the objects in the environment, and meaningful learning or understanding occurs when individuals are engaged in social activities. This philosophy is in alignment with the approach of inquiry-based science instruction, where students engage in hands-on and minds-on activities, thus placing them at the center of the learning process. In a
social constructivist classroom, rather than acting as the providers of knowledge, the role of the teacher is to provide facilitation, support and encouragement; imparting the skill of how-to-learn becomes more important than any particular information being presented. Thus, provided with proper coaching, in this type of classroom students are not receivers of knowledge but instead students construct their own understandings as they seek answers to questions (Creswell, 2009).

In recent decades, social constructivists have shifted the focus from individual learning to address collaborative and social dimensions of learning, thus bringing together the work of Jean Piaget, Jerome Bruner, and Lev Vygotsky. Both Piaget and Bruner alleged that learning occurs as a result of experience and the process of social interaction. In addition, both Bruner and Vygotsky placed language and collaborative communication, and hence instruction and learning, at the core of intellectual development (Wood, 1998). Therefore, Piaget, Bruner and Vygotsky have all substantially influenced and enhanced the collaborative development (teacher-student), group work (student-student) and overall nature of inquiry-based laboratory instruction and laboratory classroom environments.

In the last decade, chemistry educators have begun to develop and apply innovative teaching strategies adapted to the physical conditions of the learning environment founded on the social constructivists’ perspective. These include modifying lecture and laboratory activities to promote student-student group work and teacher-student collaborations, and to implement more inquiry-based science instruction in both lecture and laboratory classrooms (Gabel, 2000).

**Synthesis of the Research Literature**

Although inquiry-based instruction can be incorporated into all aspects of a science course, educational researchers and science instructors alike would agree that the most obvious choice for implementation is the laboratory classroom. Clearly, the science laboratory, if
structured correctly, provides an authentic research experience and opportunity to promote cognitive learning skills such as asking relevant questions, observing phenomena, developing critical thinking and formulating arguments in a scientific context (Hofstein, Shore & Kipnis, 2004). In addition, the construct of inquiry-based laboratory classrooms provides students with group work opportunities to collaborate, deliberate and communicate with classmates. Thus, students learn science by doing science. Research indicates that students who engage in active inquiry-based instructional and learning environments learn practical, useful approaches to solving problems and answering questions, demonstrate improved conceptual science understanding and research skills, and exhibit improved perceptions and more positive attitudes toward science (Hofstein et al., 2004).

**Guided/Open-Inquiry Laboratory Instruction**

[Inquiry] an activity consisting of three phases: (1) exploration, in which the students collect data on a system for which no theoretical background is provided; (2) invention, in which the students analyze their data and draw conclusions from it, and the excepted scientific terminology is placed on the observed behavior; (3) discovery, further student experimentation and data analysis designed to enlarge on the invented concept.

(Renner’s definition, as cited in Pavelich & Abraham, 1977, p. 24)

A four-year study by Pavelich and Abraham (1979) evaluated a guided/open-inquiry laboratory format (test group) compared to a traditional laboratory format (control group) in a two-semester sequence general chemistry course. The characteristics of three different laboratory formats are summarized in (Table 2.0) and differentiated by the nature of the information provided or not provided by the teacher. In this early study, the researchers employed a two-phase laboratory format consisting of (1) a guided-inquiry laboratory session scheduled prior to a lecture topic,
followed by (2) one or two open-inquiry laboratory sessions scheduled to coincide with the lecture topic. The two-phase laboratory format was essential for first-year chemistry students to develop the fundamental laboratory skills required to perform open-inquiry experiments.

| Table 2.0 Characteristics of Laboratory Types* (Pavelich & Abraham, 1979). |
|---------------------------------|---------------------------------|---------------------------------|
|                                 | Verification                      | Guided-Inquiry                      | Open-Inquiry                      |
| Order                           | C → D                             | D → C                             | D → C                             |
| Choice of Problem               | T                                 | T                                 | S                                 |
| Experiment Design               | T                                 | T                                 | S                                 |
| Data Analysis                   | T                                 | S                                 | S                                 |
| Data Explanation                | T                                 | S                                 | S                                 |

*C: Concepts  D: Data  T: Teacher  S: Student

Observational assessment of in-class laboratory activities was used by researchers to informally establish that both group of students developed laboratory skills, techniques and procedures equivalently. A more formal Piagetian-type paper and pencil tasks test was used to measure the effect of laboratory format on intellectual development. The tasks were administered to both student populations, at the beginning of the first-semester (~600/group), at the end of the first-semester (N = 133 for both groups) and at the end of the second-semester (N = 91 for each group). The researchers reported that after one semester the test group displayed significant gains in abstract thinking or growth in intellectual development (t = 4.71, p < 0.001); however, those results were lost after two semesters. An additional instrument, The Laboratory Program Variables Inventory (LPVI) survey, developed by the authors and validated by Abraham (1982), was used to survey the students about their laboratory experiences. The LPVI consisted of twenty-five statements concerned with various operational aspects of the laboratory formats. Students were asked to rank the statements in order, from those statements that were most descriptive of their laboratory experience, to those statements which were least descriptive. By comparing the student statement rankings, the researchers found exceptional differences between the operational aspects (e.g., the instructor lectures to the whole class) of the two laboratory
formats. Additionally, the guided/open-inquiry laboratory format was determined to be substantially better than the traditional laboratory format at encouraging scientific inquiry processes.

Inquiry-Based Laboratory Instruction

"Students work cooperatively in small groups to investigate scientific phenomenon" (Hofstein, Nahum & Shore, 2001, p. 195). Over a period of four years (1997-2000), Hofstein et al. (2001) conducted research, on 11th grade high school chemistry students using non-inquiry (control group) and inquiry-type (test group) laboratory instruction, seeking to apply a measure similar to LPVI. The Science Laboratory Environment Inventory (SLEI) survey, developed and validated by Fraser, McRobbie and Giddings (1993), was used to assess students’ perceptions of the chemistry laboratory learning environment. The SLEI consisted of sixty-eight items inside eight learning factors or scales: Student Cohesiveness, Open-Endedness, Integration, Rule Clarity, Material Environment, Teacher Supportiveness, Involvement and Organization. Student interviews and feedback questionnaires provided other sources of information regarding students’ attitudes and perceptions of the inquiry-type (test group) laboratory environment, and served as a method for validating the sensitivity of SLEI for different instructional approaches used in the science laboratory classroom. The study concluded that the students in the inquiry-type (test group) laboratory, as a result of their laboratory experiences, attained a significant improvement in perceptions of the chemistry laboratory learning environment. Students perceived that they were more involved, as measured by the Involvement scale (Mean = 3.94 vs. Mean = 3.42, t = 9.99, p < 0.000), and responsible for their own learning, Open-endedness scale (Mean = 3.27 vs. Mean = 2.20, t = 20.43, p < 0.000), and preferred the inquiry-type (test group) laboratory curriculum as compared with the non-inquiry (control group) laboratory.
Kipnis and Hofstein (2007) followed up this study with a direct comparison of inquiry-type and traditional chemistry laboratory environments over a period of six years (1999-2005). The researchers expanded their study to over 3500 high school chemistry students and added a practical test to assess the development of inquiry skills as well as an *Attitude Towards Science Laboratory* (ATSL) questionnaire to capture student attitude towards laboratory work. The ATSL questionnaire, developed and validated by Hofstein, Ben-Zvi and Samual (1976), included sixty-two items that were divided into eight factors or scales: (I) Learning in the chemistry lab, (II) Amount of laboratory work, (III) Value of laboratory work, (IV) The place of laboratory work within the framework of chemistry teaching and its value as a mean for learning chemistry, (V) Students’ enthusiasm for practical work and their enjoyment in working in the lab, (VI) Students’ assessment of their own experimentation versus teacher’s demonstration, (VII) Immediate and future benefits students gain from experimentation, and (VIII) The advantage of laboratory work. This measure was administered at the start of the 11th grade and the end of the 12th grade. The practical test revealed that students in the inquiry-type (test group) curriculum asked significantly more questions (Mean = 5.19 vs. Mean = 3.05, t = 10.55, p ≤ 0.0001) and that those questions were more high-level, quantitative inquiry-type questions than the non-inquiry (control group), suggesting that the teaching technique had a positive impact on the development of inquiry skills. Analysis of the SLEI survey yielded similar results as the Hofstein et al. (2001) study. Analysis of the ATSL questionnaire revealed no significant difference between the inquiry-type (test group) and non-inquiry (control group) in the 11th grade. However, the comparison conducted at the end of the 12th grade revealed the inquiry-type (test group) demonstrated a more positive attitude towards chemistry laboratory work compared to the non-inquiry (control group) on five (III, IV, V, VI and VII) dimensions of attitude. The researchers
concluded that the inquiry-type program provided the students with the opportunity to be involved in a worthwhile (student-centered) learning process that the chemistry laboratory provided.

**Inquiry-Discovery Laboratory Instruction**

"The experiments used in this study were written to assist the student in discovering some of the important laws of chemistry for himself. The experiments utilized the inquiry approach to learning" (Richardson & Renner, 1970, p. 77). A three-year study led by Richardson and Renner (1970) used control and experimental student groups measured against each other to investigate the effects of inquiry-discovery laboratory instruction on student laboratory final exams during three consecutive years (Fall 1966, Fall 1967 and Spring 1968) and laboratory pre- and post-tests (Spring 1968) for the beginning college chemistry laboratory. The experimental variables became more controlled as the study progressed, yielding the third-year (Spring 1968), with the most consistent experimental controls. For example, the third-year controls involved (1) one experimental group and one control group, (2) the eight laboratory experiments performed by each group were matched for content, (3) eight pre- and post-tests were administered at the time each experiment was performed, (4) both groups of students had the same lab instructor, lecture instructor and lab assistant. Therefore, the only experimental variable was the difference in laboratory format, thus the data set from 1968 was considered the most reliable. Likewise, the authors stated "the superiority of the inquiry method of laboratory instruction over that of the conventional method is probably best provided by the interpretation of the data collected during the Spring semester 1968" (Richardson & Renner, 1970, p. 78).

With that said, the authors reported that the (experimental group) students in the inquiry-discovery laboratory performed significantly better on the final laboratory exam for all three
semesters when compared to the (control group) students in the conventional laboratory (Table 2.1).

**Table 2.1 Final Laboratory Exam Mean Scores by Laboratory Method.**

<table>
<thead>
<tr>
<th>Year (Semester)</th>
<th>Verification Lab (Control Group)</th>
<th>Inquiry-Discovery Lab (Experimental Group)</th>
</tr>
</thead>
<tbody>
<tr>
<td>1966 (Fall)</td>
<td>*20.90</td>
<td>*25.90</td>
</tr>
<tr>
<td>1967 (Fall)</td>
<td>*17.06</td>
<td>*20.48a, *21.00b</td>
</tr>
<tr>
<td>1968 (Spring)</td>
<td>*18.44</td>
<td>*21.58</td>
</tr>
</tbody>
</table>

*p < 0.05, a: first independent experimental group, b: second independent experimental group

In addition, statistical analysis of the pre- and post-test data (Spring 1968) showed that the inquiry-discovery laboratory students (experimental group) achieved significantly better (at the 0.05 level of significance) over the conventional laboratory students (control group) on all eight of the pre- and post-tests.

The next study describes inquiry-discovery laboratory instruction as “a sequence of events designed by the students under the guidance of a teacher in which students construct meaningful knowledge about phenomena in the laboratory setting” (Bodner, Hunter & Lamba, 1998. p. 6). In a extensive implementation of inquiry-based discovery laboratories into the first half of a two-semester general chemistry curriculum, Bodner et al. (1998) utilized an Action Research method where all of the students (~400), working in the laboratory in groups of three, were treated to the same intervention. The researchers used laboratory observations and interviews to report that both students and teaching assistants expressed overall contentment with their laboratory experiences despite frustrations caused by the non-prescriptive nature of the inquiry-discovery laboratory format. Additionally, the use of a Likert-scale survey indicated that 86% of the students would recommend or strongly recommend the inquiry instruction in the laboratory and 74% either agreed or strongly agreed that working in groups helped them
understand course material, implying that the inquiry-discovery teaching approach had a significant and favorable effect on student attitude and disposition.

**Inquiry-Based, Cooperative Learning, Laboratory Instruction**

"Laboratory format for general chemistry that exposes students to the process of scientific problem solving, emphasizes collaborative work, and requires the students to communicate their results both orally and in writing" (Cooper, 1994, p. 307). In a large-scale general chemistry experiment conducted by Cooper (1994) involving 2000 students per semester, student surveys indicated that the inquiry-based cooperative learning (4 students per group) laboratory setting offered a more positive laboratory experience, and the students believed they learned more from their laboratory experience. Cooper (1994) also indicated that the combination of cooperative learning inside an inquiry laboratory curriculum impacted student performance in lecture. All students, regardless of laboratory type, take the same lecture exams; therefore analysis of lecture grades showed a gender difference and revealed that female students enrolled in the inquiry-based cooperative learning (test group) laboratory curriculum outperformed their female counterparts enrolled in the traditional (control group) laboratory curriculum by as much as 10%. Despite the positive outcome shown for female students, a similar correlation was not found for male students. Moreover, an examination of drop rates also corroborated a gender difference. The drop rate for females in the inquiry-based cooperative learning (test group) laboratory was found to be 13% compared to 21% for the traditional (control group) laboratory and male students dropped the course 9% regardless of laboratory type.

"[Guided-inquiry] experiments are designed to lead students to hypothesis formation and testing. This approach is based on the learning cycle, which consists of three phases: data
collection, concept invention, and application” (Farrell, Moog & Spencer, 1999, p. 572). A four-year study by Farrell et al. (1999) presents a case study with respect to the implementation of guided-inquiry laboratory instruction in a two-semester sequence general chemistry course for science majors. The researchers evaluated final exams, course grades, which consisted of quizzes (10% of overall course grade), group participation (5% of overall course grade), lecture exams (65% of the overall course grade) and laboratory reports (20% of the overall course grade), and withdrawal grades (W-grades) as performance measures to investigate the effects of the guided-inquiry cooperative learning curriculum on student achievement. The authors reported a grade distribution (Table 2.2) that demonstrates an increase in ‘A’ (4.9%) and ‘B’ (7.5%) overall course grades, a decrease in ‘D’ (1.9%) and ‘F’ (3.4%) overall course grades and a decrease in course withdrawal ‘W’ (7.0%) grades for students enrolled in the guided-inquiry laboratory curriculum. In addition, the final exams of the guided-inquiry (test group) students scored as high as or higher than the students in the traditional (control group) curriculum. These results indicated that the percentage of students successfully completing the chemistry course substantially increased (12.3%) as a result of the new approach. In addition, the authors stated that they intended to continue using the guided-inquiry cooperative learning instructional technique because they did not discover any negative attributes that would lead them to abandon it.

Table 2.2 Grade Distribution for Authors’ Sections of General Chemistry (Farrell et al., 1999).

<table>
<thead>
<tr>
<th>Period</th>
<th>Curriculum</th>
<th>Percentage of Students Earning Grade</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>A</td>
</tr>
<tr>
<td>F90-S94</td>
<td>Traditional</td>
<td>19.3</td>
</tr>
<tr>
<td>F94-F97</td>
<td>Guided Inquiry</td>
<td>24.2</td>
</tr>
</tbody>
</table>

28
Guided-Inquiry, Collaborative, Science Writing Heuristic (SWH), Laboratory Instruction

“When using SWH, the role of the instructor changes. [The instructor] serves as a facilitator and helps guide students to design an experiment to answer the student’s questions” (Greenbowe & Hand, 2005). A study by Greenbowe and Hand (2005) investigated the effects of a guided-inquiry, group work laboratory format in conjunction with the SWS laboratory report writing technique in the first-semester general chemistry by comparing American Chemical Society (ACS) standardized exam scores administered at the beginning of the semester against ACS exam scores administered at the end of the semester. At the beginning of the semester, there was a significant difference ($F(1, 285) = 14.5298, p < 0.001$) in favor of male students (62.7%), over female students (56.5%), on the American Chemical Society (ACS) California Diagnostic Test scores; however, by the end of the semester, there was no significant difference ($F(1, 236) = 0.0822, p = 0.775$) between male students (78.3%) and female students (76.5%) on the ACS First-Semester General Chemistry Examination scores. The study indicated that female students made significant improvements in their level of chemistry knowledge, and substantial gains in the gender gap, which was reduced from 0.45 at the beginning of the semester to 0.04 at the end of the semester.

Poock, Burke, Greenbowe and Hand (2007) extended this study to encompass a two-semester longitudinal study involving 78 science and engineering students during their first-semester and second-semester in both the lecture and laboratory segments of a general chemistry course. The researchers reported a clear statistically significant difference ($F = 4.298, p = 0.0074$) in students’ lecture academic performance (average total points in the lecture portion of the course) at the end of the first-semester between students (Mean = 83.1%) having a teaching assistant (TA) rated high in implementing the SWH practices in the laboratory compared to
students (Mean = 74.8%) having a TA rated low. Similarly the researchers reported a statistically significant difference ($F = 6.071, p = 0.0010$) in students' lecture academic performance (average total points in the lecture portion of the course) at the end of the second-semester between students (Mean = 83.7%) having a teaching assistant (TA) rated high in implementing the SWH practices in the laboratory compared to students (Mean = 67.8%) having a TA rated low. The researchers concluded that students benefit when an instructor or TA proficiently utilizes the guided-inquiry, collaborative, SWH laboratory approach and engages their students in the laboratory portion of the course.

Summary

An account of the research study (year), a brief description of the type of chemistry course (interval of study), the featured laboratory format (group size), performance measure data collection methods (E: exams, I: interviews, O: observations, P: pre-/post-test &/or practical test, V: surveys, Q: questionnaires) and results in terms of student outcomes with respect to each research study considered in the previous section is summarized and presented below in chronological order (Table 2.3).

Table 2.3 Summary of Research Literature Results.

<table>
<thead>
<tr>
<th>Research Study (year)</th>
<th>Course Type</th>
<th>Laboratory Format (group size)</th>
<th>Performance Measure(s)</th>
<th>Student Outcomes</th>
</tr>
</thead>
<tbody>
<tr>
<td>Richardson &amp; Renner (1970)</td>
<td>Beginning College Chemistry</td>
<td>Inquiry-Discovery</td>
<td>X, X</td>
<td>Significantly ↑ students' scores on lab final exam &amp; pre-/post-tests</td>
</tr>
<tr>
<td>Pavalich &amp; Abraham (1979)</td>
<td>General Chemistry (1st year)</td>
<td>Guided/Open-Inquiry (student pairs)</td>
<td>X, X, X</td>
<td>Significant gain in intellectual development &amp; substantially better at encouraging scientific inquiry processes</td>
</tr>
</tbody>
</table>

30
<table>
<thead>
<tr>
<th>Research Study (year)</th>
<th>Course Type</th>
<th>Laboratory Format (group size)</th>
<th>Performance Measure(s)</th>
<th>Student Outcomes</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cooper (1994)</td>
<td>General Chemistry (1 semester)</td>
<td>Inquiry-Based Cooperative Learning (4 students)</td>
<td>X</td>
<td>↑ (+) Attitudes, ↑ lecture exam &amp; ↑ retention rates (particularly for female students)</td>
</tr>
<tr>
<td>Bodner et al. (1998)</td>
<td>General Chemistry (1st semester)</td>
<td>Inquiry-Discovery (3 students)</td>
<td>X X X</td>
<td>Inspired (+) lab climates &amp; (+) student attitudes</td>
</tr>
<tr>
<td>Farrell et al. (1999)</td>
<td>General Chemistry (1st semester)</td>
<td>Guided-Inquiry Discovery (4 students)</td>
<td>X</td>
<td>↑ Final course grades &amp; ↓ withdrawal rates = ↑ students success completing the course</td>
</tr>
<tr>
<td>Hofstein et al. (2001)</td>
<td>High School Chemistry (11th grade)</td>
<td>Inquiry-Type (3-4 students)</td>
<td>X X X</td>
<td>Students preferred the inquiry-type laboratory &amp; perceptions of the lab significantly improved</td>
</tr>
<tr>
<td>Greenbowe &amp; Hand (2005)</td>
<td>General Chemistry (1st semester)</td>
<td>Guided-Inquiry, Collaborative, SWH</td>
<td>X</td>
<td>Female students made significant improvements in their level of chemistry knowledge</td>
</tr>
<tr>
<td>Kipnis &amp; Hofstein (2007)</td>
<td>High School Chemistry (11th &amp; 12th grades)</td>
<td>Inquiry-Type (3-4 students)</td>
<td>X X X</td>
<td>Students developed inquiry skills &amp; a more (+) attitude toward chemistry work in 12th grade</td>
</tr>
<tr>
<td>Poock et al. (2007)</td>
<td>General Chemistry (1st year, 2 semesters)</td>
<td>Guided-Inquiry, Collaborative, SWH</td>
<td>X X</td>
<td>↑ Students’ mean total points in the lecture portion of the course for both 1st and 2nd semester</td>
</tr>
</tbody>
</table>

Instruments (E: Exams, I: Interviews, O: Observations, P: Pre-/Post-test &/or Practical Test, Q: Questionnaires, V: Surveys)

**Conclusion**

The literature review presents several research studies (quantitative and qualitative) that were conducted on the effectiveness of inquiry-based laboratory instruction in a variety of
chemistry (high school and general chemistry) laboratory courses. However, it is problematic to compare these studies because (1) the definition of inquiry-based laboratory instruction is extensive and lacks consistency; (2) these studies used several different performance measures (exams, interviews, observations, tests, questionnaires and surveys); to (3) assess many different student outcomes (improvement in student intellectual gains, attitudes, encouraging inquiry, laboratory perceptions and student achievement). Furthermore, it is most difficult to compare these studies because the degree or level of inquiry present in the laboratory experiences was not evaluated and characterized. Hence, this thesis will add to the literature base by evaluating the effect of inquiry-based laboratory instruction as a function of the level of inquiry present in the new guided-inquiry laboratory curriculum.
Chapter Three: Research Design

Introduction

Chapter Three considers the research question, “How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at GRCC?” and addresses a variety of essential research design features. As a result, this chapter (1) provides a description of the two student test groups, (2) considers the method used to establish the reliability of the modified for college instruction rubric, (3) discusses the data collection and statistical analysis techniques, and (4) concludes with a brief summary.

Participants/Subjects

A quasi-experimental design and nonrandom sampling procedure was used to establish two student test groups from intact laboratory sections, three semesters from the old traditional laboratory curriculum (Fall 2006, Winter 2007 and Fall 2007) and three semesters from the new guided-inquiry laboratory curriculum (Fall 2008, Winter 2009 and Fall 2009). In order to minimize instructor effect, data was collected and provided by the same laboratory instructor throughout the data collection segment of the study for all six semesters.

ACT test scores of the two student test groups were compared to establish population equivalency. There were a total of 183 students with allocated laboratory grades in the original data set. ACT scores were only available for 128 of these students. This smaller, n = 128, sample was used for statistical analysis. The control group consisted of 58 students who performed traditional verification laboratory experiments, while the experimental group consisted of 70 students who performed the guided-inquiry laboratory experiments.
Instrumentation: The Continuum of Scientific Inquiry Rubric

The Continuum of Scientific Inquiry Rubric: Inter-Rater Reliability

The researchers in this study used the modified rubric for college instruction advanced by Fay et al. (2007) to characterize the level of inquiry present in each laboratory experiment and both laboratory curricula overall. As such, Fay et al. (2007) established the dependability of the modified for college instruction Continuum of Scientific Inquiry Rubric (Table 3.0) to distinguish between levels of inquiry present in a wide spectrum of chemistry laboratory activities covering a comprehensive selection of chemistry concepts. Included within the selection of chemistry concepts were chemically similar activities as well. Fay et al. (2007) commissioned a team of three researchers to evaluate all 27 individual chemistry experiments section-by-section: (1) pre-lab, (2) procedure, and (3) post-lab calculations and conclusions, across 12 different resources, twice over. The first set of 18 general chemistry/environmental chemistry laboratory experiments were chosen from three commercially published laboratory curricula and the second set consisted of 9 organic chemistry laboratory experiments that were selected from the Journal of Chemical Education by searching the terms ‘inquiry’ and ‘discovery-based learning.’ Upon completion of the second evaluation by researchers, an inter-rater reliability (IRR) value was calculated separately for each category of experiments (general/environmental chemistry = 18 individual experiments = 0.89), (organic chemistry = 9 individual experiments = 0.78) and overall for the entire collection of all 27 experiments (overall = 0.85). Since the lowest acceptable value for establishing reliability is 0.70, the high overall 0.85 IRR rendered the modified rubric as robust and the reliability of the modified rubric to characterize the level of inquiry present in college chemistry laboratory experiments and overall laboratory curricula is considered good (Fay, et al., 2007).
Table 3.0: The Continuum of Scientific Inquiry Rubric (Fay et al., 2007).

<table>
<thead>
<tr>
<th>Level of Inquiry</th>
<th>Description</th>
</tr>
</thead>
<tbody>
<tr>
<td>Level-0</td>
<td>The problem, procedure and methods to solutions are provided to the student. The student performs the experiment and verifies the results with the manual.</td>
</tr>
<tr>
<td>Level-1</td>
<td>The problem and procedure are provided to the student. The student interprets the data in order to propose viable solutions.</td>
</tr>
<tr>
<td>Level-2</td>
<td>The problem is provided to the student. The student develops a procedure for investigating the problem, decides what data to gather, and interprets the data in order to propose viable solutions.</td>
</tr>
<tr>
<td>Level-3</td>
<td>A “raw” phenomenon is provided to the student. The student chooses the problem to explore, develops a procedure for investigating the problem, decides what data to gather, and interprets the data in order to propose viable solutions.</td>
</tr>
</tbody>
</table>

**Data Collection and Statistical Analysis**

**Inter-Rater Reliability and Level(s) of Inquiry**

Given the robust nature and good reliability of the modified rubric (Table 3.0), the level of inquiry present in the 14 laboratory experiments from both of GRCC’s general chemistry laboratory curricula (traditional and guided-inquiry) were analyzed and characterized independently by two researchers, and an IRR was calculated. Laboratory experiments comprised of multiple sections were assigned a single inquiry-level, which was determined by the highest level of inquiry present in any of the sections of the experiment. A Spearman’s Rho (Field, 2005) correlation coefficient was used to assess the inter-rater reliability of the level of inquiry classification for each (14 experiments) laboratory curricula and overall (28 experiments). A graphic distribution of experiments with respect to levels of inquiry across laboratory curricula was developed.

A statistical comparison of the levels of inquiry in the two curricula was done using a Mann-Whitney (U) test (Field, 2005), which is the non-parametric equivalent of a t-test. This more conservative comparison was warranted because the inquiry levels were not normally distributed.
Student Performance: Effect of Curriculum

The first student performance data set consisted of students’ overall laboratory report average final grades from intact laboratory sections, three semesters from the old traditional laboratory curriculum (Fall 2006, Winter 2007 and Fall 2007) and three semesters from the new guided-inquiry laboratory curriculum (Fall 2008, Winter 2009 and Fall 2009). Initially, the student performance evaluation compared the Laboratory Report Average Final Grade (LRAFG) and an Adjusted Laboratory Report Average Final Grade (ALRAFG) of both student groups for both laboratory curricula. The LRAFG was calculated by using all of the 14 possible laboratory report scores per semester and included any/all zero scores assessed for laboratory reports that were not submitted to the instructor for grading. The ALRAFG was derived solely from those laboratory reports that were completed and submitted to the instructor for grading, thus the ALRAFG excluded any/all laboratory report scores of zero. The LRAFG and ALRAFG for both student test groups were statistically compared using a Mann-Whitney (U) test, because the grades were not normally distributed.

Student Performance: Effect of Level of Inquiry

The second student performance data set consisted of students’ individual laboratory report scores from intact laboratory sections, three semesters from the old traditional laboratory curriculum (Fall 2006, Winter 2007 and Fall 2007) and three semesters from the new guided-inquiry laboratory curriculum (Fall 2008, Winter 2009 and Fall 2009). This performance evaluation pooled all students into one group, omitted the type of laboratory curriculum, and compared individual laboratory report scores by the level of inquiry assigned to the laboratory experiment. For ease of comparison, the scores for each laboratory report were converted to a 10.00-point scale. Ideally, had every student submitted each laboratory report for instructor
grading, there would have been a total of 1792 individual laboratory report scores produced. However, scores were only available for 1564 laboratory reports; consequently the comparison was made using these. The laboratory report scores were not normally distributed so the comparison was done using the Kruskal-Wallis (H) test (Field, 2005), a non-parametric Analysis of Variance. In addition, Mann-Whitney (U) tests with a Bonferroni correction (Field, 2005) were used to determine which levels of inquiry (Level-1, Level-2 and/or Level-3) were different from the others.

**Student Perseverance**

The student perseverance data set consisted of completed student laboratory reports and student withdrawal grades (W-grades) from intact laboratory sections, three semesters from the old traditional laboratory curriculum (Fall 2006, Winter 2007 and Fall 2007) and three semesters from the new guided-inquiry laboratory curriculum (Fall 2008, Winter 2009 and Fall 2009). The student perseverance evaluation compared the percentage of completed laboratory reports and the percentage of withdrawal grades of both student test groups for both laboratory curricula. The comparisons were made via Mann-Whitney (U) tests, because the data were not normally distributed. In addition, a subgroup statistical analysis of student perseverance for students that did not complete all fourteen laboratory reports was used to capture the effect of curriculum on students that might be in danger of not completing the course.

**Summary**

A quasi-experimental design and nonrandom sampling procedure was used to establish two student test groups from intact laboratory sections. The work of Fay et al. (2007) provided a robust measuring instrument and a reliable rubric to use for the purpose of assessing and establishing the level(s) of inquiry present in college level chemistry laboratory activities,
experiments and curriculum. Thus, this study used the modified for college instruction version of The Continuum of Scientific Inquiry Rubric (Fay et al., 2007) to assess and establish the level of inquiry integrated into each laboratory experiment and both laboratory curricula overall.

In addition, instructor effect was minimized by analyzing only those data that were collected and provided by the same laboratory instructor throughout the data collection segment of the study for all six semesters. Furthermore, student performance and perseverance assessments feature the creation of a product (laboratory reports) on an ordinal scale of unequal scale intervals. Student performance was evaluated and statistically analyzed via laboratory report average final grades (LRAFG) and individual laboratory report scores, whereas student perseverance was measured and statistically analyzed by comparing the overall percentage of completion rates for laboratory reports by curriculum to ascertain if any relationships exist between level(s) of inquiry and student performance and student perseverance in the laboratory portion of the course.
Chapter Four: Results

Context

This study was conducted at GRCC, a two-year community college located in the Great Lakes region of the United States. GRCC practices an open admission policy and serves 30,000 students annually. The student population involved in this research, n = 128, was composed of students enrolled in CM103, which is the first sequence of a two-semester general chemistry course intended to educate students majoring in science and engineering programs. CM103 was a five-credit course comprised of 4 hours of lecture (~70-80% of course grade) and three hours of laboratory (~20-30% of course grade) each week. The laboratory component of CM103 was taught by the same chemistry laboratory instructor throughout the data collection segment of this research and featured a pre-laboratory quiz (~15 minutes), a pre-laboratory lecture (~15 minutes) followed by a laboratory exercise (~2.5 hours). In the laboratory classroom students enrolled in the traditional laboratory curriculum (Fall semester 2006 - Fall semester 2007) worked in pairs, while the students enrolled in the inquiry-based laboratory curriculum (Fall semester 2008 – Fall semester 2009) worked in groups of (2-4 students). Each student was required to write his or her own laboratory report for each laboratory exercise that was completed in both laboratory curricula. Laboratory reports that were submitted to the instructor for grading during the traditional laboratory curriculum were scored on a 10.00-point scale. However, laboratory reports submitted to the instructor for grading during the inquiry-based laboratory curriculum were scored on a 15.00-point scale and later converted to a 10.00-point scale for comparison via statistical analysis.
Findings

In an effort to answer the research question, “How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at GRCC?” and address the hypothesis, “The implementation of a more inquiry-based instructional laboratory curriculum will positively impact student performance and student perseverance in the laboratory portion of the first-semester general chemistry course (CM103) designed to educate students majoring in science and engineering programs” statistical analysis of data took many forms. As a result, this chapter (1) describes the analysis of students’ ACT scores used to establish that the two student test groups were from the same population, (2) considers the level(s) of inquiry present in laboratory experiments and curricula, (3) discusses the student performance and perseverance data, and (4) concludes with a summary of the results.

Participants/Subjects

A comparison of mean ACT scores for each test group of students (Table 4.0) shows that the scores were normally distributed and have equivalent variances, both of which are requirements for comparison of the groups using a t-test. On average, there was no difference in the ACT comprehensive score of the students in the traditional laboratory curriculum and the guided-inquiry laboratory curriculum, t(126) = -0.12, p > 0.05. The same was true for the students’ ACT science scores, t(126) = -0.21, p > 0.05. Therefore, the two student test groups, those completing the old traditional verification laboratory activities and those completing the new guided-inquiry laboratory activities, were treated as equivalent samples drawn from the same population permitting statistical comparison of student performance and student perseverance.
Table 4.0 Comparison of Average ACT Scores by Student Group.

<table>
<thead>
<tr>
<th>Students (Curriculum Type)</th>
<th>Mean ACT Comp. Score (SD)</th>
<th>Mean ACT Science Score (SD)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control, N = 58 (Traditional Laboratory)</td>
<td>21.2 (4.0)</td>
<td>22.2 (3.7)</td>
</tr>
<tr>
<td>Experimental, N = 70 (Guided-Inquiry Laboratory)</td>
<td>21.3 (3.4)</td>
<td>22.3 (3.3)</td>
</tr>
</tbody>
</table>

Significant, $p < 0.05$

Inter-Rater Reliability and Level(s) of Inquiry

To establish the level of inquiry present in each laboratory experiment and laboratory curriculum, the modified rubric for college instruction described in Chapter’s One (Table 1.0) and Three (Table 3.0) was used by two evaluators. A Spearman’s Rho correlation coefficient was used to establish inter-rater reliability (Table 4.1). Given the fairly strong agreement, indicated by the IRR value ($0.85, p < 0.01$), the level of inquiry assigned to each experiment in both laboratory curricula was determined (Table 4.2). When the initial level of inquiry assessed for any given experiment resulted in disagreement, discussions between the two researchers negotiated the final level of inquiry assigned to that experiment. Furthermore, it’s important to note that any initial disagreement with respect to an assessed inquiry-level between the two researchers was never more than one level away from each other.

Table 4.1 Inter-Rater Reliability Values for Laboratory Curricula.

<table>
<thead>
<tr>
<th>Laboratory Curriculum</th>
<th>Number of Experiments with Agreement</th>
<th>Total Number of Experiments</th>
<th>Inter-Rater Reliability (Spearman’s Rho)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Traditional</td>
<td>12</td>
<td>14</td>
<td>0.73*</td>
</tr>
<tr>
<td>Guided-Inquiry</td>
<td>12</td>
<td>14</td>
<td>0.74*</td>
</tr>
<tr>
<td>Overall</td>
<td>24</td>
<td>28</td>
<td>0.85*</td>
</tr>
</tbody>
</table>

*Reliability significant, $p < 0.01$
### Table 4.2 Laboratory Curricula: Level of Inquiry.

<table>
<thead>
<tr>
<th>Experiment: Traditional Laboratory Curriculum (Chemistry Concepts)</th>
<th>Level of Inquiry</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Introduction to the Chemistry Lab</td>
<td>0</td>
</tr>
<tr>
<td>2. <strong>Determination of Density</strong></td>
<td>0</td>
</tr>
<tr>
<td>3. Separation of a Mixture (extraction using acid/base properties)</td>
<td>0</td>
</tr>
<tr>
<td>4. Electrons &amp; Light (atomic spectra &amp; unknown identification)</td>
<td>1</td>
</tr>
<tr>
<td>5. Chemicals &amp; Reactions (observe, describe &amp; identify chemical changes)</td>
<td>1</td>
</tr>
<tr>
<td>6. Solution Chemistry (concentration of a solution determination)</td>
<td>0</td>
</tr>
<tr>
<td>7. Molecular Modeling (Lewis dot structures &amp; molecular geometries)</td>
<td>0</td>
</tr>
<tr>
<td>8. Separation of Compounds (intermolecular forces)</td>
<td>1</td>
</tr>
<tr>
<td>9. A Cycle of Copper Reactions (reaction types)</td>
<td>0</td>
</tr>
<tr>
<td>10. Stoichiometry</td>
<td>0</td>
</tr>
<tr>
<td>11. Determining Concentration (Beer's law)</td>
<td>0</td>
</tr>
<tr>
<td>12. Enthalpy Changes &amp; Hess’s Law (calorimetry-heat of solution)</td>
<td>0</td>
</tr>
<tr>
<td>13. <strong>Signs of ΔH, ΔS &amp; ΔG</strong> (qualitative thermodynamics)</td>
<td>1</td>
</tr>
<tr>
<td>14. Ten Unknowns (experimental design &amp; unknown identification)</td>
<td>1</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Experiment: Guided-Inquiry Laboratory Curriculum (Concepts)</th>
<th>Level of Inquiry</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Introduction to the Chemistry Lab.</td>
<td>0</td>
</tr>
<tr>
<td>2. How Long Can a Bubble Last? (experimental design)</td>
<td>2</td>
</tr>
<tr>
<td>3. Are All Pennies the Same? (density)</td>
<td>1</td>
</tr>
<tr>
<td>4. What Makes a Solution Colored or Colorless? (absorbance, color &amp; periodic trends)</td>
<td>1</td>
</tr>
<tr>
<td>5. What Factors Affect Color Intensity? (solution prep &amp; Beer’s law)</td>
<td>1</td>
</tr>
<tr>
<td>6. Are There Property Patterns? (reactivity &amp; periodic properties)</td>
<td>1</td>
</tr>
<tr>
<td>7. What Factors Affect the Solubility of Ions? (double replacement reactions)</td>
<td>2</td>
</tr>
<tr>
<td>8. Can Toxic Ions be Removed from Water by Precipitation? (separation of compounds by solubility)</td>
<td>2</td>
</tr>
<tr>
<td>9. <strong>Do Like Repel or Attract?</strong> (intermolecular forces &amp; solubility)</td>
<td>1</td>
</tr>
<tr>
<td>10. How Much Hydrogen? (stoichiometry &amp; reactivity of metals with acid)</td>
<td>1</td>
</tr>
<tr>
<td>11. How Much is Too Much? (stoichiometry &amp; limiting reagents)</td>
<td>1</td>
</tr>
<tr>
<td>12. Which Salts Make Good Cold Packs &amp; Hot Packs? (calorimetry-heat of solution)</td>
<td>1</td>
</tr>
<tr>
<td>13. How is Heat Measured Indirectly? (calorimetry-Hess’s law)</td>
<td>1</td>
</tr>
<tr>
<td>14. <strong>Thermodynamic Signs</strong> (qualitative thermodynamics)</td>
<td>1</td>
</tr>
</tbody>
</table>

**Bold italicized text denotes initial disagreement, and the value posted is the agreed-on level.**

A distribution of the 14 laboratory experiments with respect to levels of inquiry across each laboratory curriculum was developed (Figure 4.0). The majority of the experiments in the traditional laboratory curriculum were rated Level-0 (64.3%) with the remainder rated Level-1.
(35.7%); conversely, the experiments in the guided-inquiry laboratory curriculum were rated Level-0 (7.1%), Level-1 (71.4%) and Level-2 (21.4%), respectively.

**Figure 4.0 Distribution of Levels of Inquiry Across Laboratory Curriculum.**

<table>
<thead>
<tr>
<th>Traditional Lab Curriculum: Level of Inquiry</th>
<th>Inquiry-Based Lab Curriculum: Level of Inquiry</th>
</tr>
</thead>
<tbody>
<tr>
<td>Level 0 64.3%</td>
<td>Level 0 7.1%</td>
</tr>
<tr>
<td>Level 1 35.7%</td>
<td>Level 1 71.4%</td>
</tr>
<tr>
<td></td>
<td>Level 2 21.4%</td>
</tr>
</tbody>
</table>

A Mann-Whitney (U) test, found that the level of inquiry in the guided-inquiry curriculum (Median = Level-1) was significantly higher than that of the traditional curriculum (Median = Level-0), $U = 34.50, p < 0.005, r = -0.62$.

Before comparing student performance, it is important to note that the new inquiry-based laboratory experiments were determined to be suitable replacements for the old traditional laboratory experiments by the chemistry department at GRCC. Therefore, the inquiry-based laboratory experiments should have a comparable level of course rigor and expectations of the students.

**Student Performance: Effect of Curriculum**

The LRAFG and ALRAFG for both student test groups were statistically compared (Table 4.3) using the Mann-Whitney (U) test because they were not normally distributed. Data analysis revealed that the LRAFG for the students in the guided-inquiry curriculum (Median = 85.00%) was not significantly lower than that of the students in the traditional curriculum (Median = 89.93%). On the other hand, the ALRAFG for the students in the guided-inquiry
curriculum (Median = 90.07%) was significantly lower than that of the students in the traditional curriculum (Median = 94.74%), $U = 925.00, p < 0.005, r = -0.47$.

**Table 4.3 Comparisons of LRAFG and ALRAFG by Student Group.**

<table>
<thead>
<tr>
<th>Student Test Group (Lab Curriculum)</th>
<th>Median/Mean LRAFG (SD) [%]</th>
<th>Median/Mean ALRAFG (SD) [%]</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control, N = 58 (Traditional)</td>
<td>89.93/77.59 (27.25)</td>
<td>94.74/93.89 (3.83)</td>
</tr>
<tr>
<td>Experimental, N = 70 (Guided-Inquiry)</td>
<td>85.00/81.00 (15.29)</td>
<td>90.07/88.75 (7.01)</td>
</tr>
</tbody>
</table>

*Note: LRAFG= laboratory report average final grade; ALRAFG= adjusted laboratory report average final grade*

Given that both types of laboratory curricula were instructed and graded by the same instructor, and covered similar topics, these data may indicate that expectations of the students in the guided-inquiry curriculum were greater and that the inquiry laboratory reports might be better probes of student understanding than the traditional laboratory reports.

**Student Performance: Effect of Level of Inquiry**

A Kruskal-Wallis (H) test, a non-parametric equivalent of an analysis of variance, revealed a significant effect with respect to level of inquiry on the individual laboratory report scores, $H(2) = 92.34, p < 0.005$. Therefore, Mann-Whitney (U) tests with a Bonferroni correction were used to determine which level of inquiry was different from the others.

Statistical analysis determined that the laboratory report scores from experiments allocated as Level-1 (Median = 9.30) or Level-2 (Median = 9.20) were not significantly different (Table 4.4). On the other hand, laboratory report scores from experiments allocated as Level-0 (Median = 9.50) were statistically higher than those scores from experiments rated as Level-1 (Median = 9.30), $U = 153487.00, p < 0.005, r = -0.25$ and as Level-2 (Median = 9.20), $U = 34933.50, p < 0.005, r = -0.25$, respectively.
Table 4.4 Comparison of Individual Laboratory Report Score by Level of Inquiry.

<table>
<thead>
<tr>
<th>Level of Inquiry</th>
<th>Number of Laboratory Reports</th>
<th>Median/Mean Laboratory Report Score (SD) [10.00-Point Scale]</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>509</td>
<td>9.50/9.39 (0.84)</td>
</tr>
<tr>
<td>1</td>
<td>855</td>
<td>9.30/8.99 (1.04)</td>
</tr>
<tr>
<td>2</td>
<td>200</td>
<td>9.20/8.86 (1.27)</td>
</tr>
</tbody>
</table>

Note: Level-0 was determined to be significantly higher than Level-1 and Level-2

Therefore, the data suggests that the inclusion of inquiry, at any level, into a laboratory experiment increases the degree of difficulty, which in turn yields lower laboratory report scores resulting in lower laboratory report average final grades. This is a reasonable finding since one goal of inquiry instruction is to enhance the development of critical-thinking skills, a process that requires effort.

Student Perseverance

The traditional laboratory curriculum produced 672 completed laboratory reports out of a possible 812, resulting in a completion rate of 82.8%. Additionally, the guided-inquiry curriculum yielded 892 completed laboratory reports out of a possible 980, which resulted in a completion rate of 91.0%. While these results appear different, statistical analysis via a Mann-Whitney (U) test indicated that the results were not statistically different (Table 4.5).

Table 4.5 Student Perseverance.

<table>
<thead>
<tr>
<th>Curriculum (Semester/Year)</th>
<th>Laboratory Report Completion Rate</th>
<th>Number of Lab Reports Submitted by Subgroup (Median)</th>
<th>Withdrawal Rate (Median)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Traditional (F06, W07, F07)</td>
<td>82.8%</td>
<td>9.5</td>
<td>18.8%</td>
</tr>
<tr>
<td>Guided-Inquiry (F08, W09, F09)</td>
<td>91.0%</td>
<td>12.0</td>
<td>6.1%</td>
</tr>
<tr>
<td>College-wide (F06, W07, F07, F08, W09, F09)</td>
<td></td>
<td></td>
<td>Mean = 14.1%</td>
</tr>
</tbody>
</table>

Note: Subgroup refers to only those students who submitted less than 14 laboratory reports

However, a Mann-Whitney (U) test of the subgroup, containing only those students who submitted less than the 14 possible laboratory reports for a grade, revealed that the subgroup of
students in the guided-inquiry curriculum submitted significantly more laboratory reports (Median = 12.00) than the subgroup of students in the traditional curriculum (Median = 9.50), U = 226.00, p < 0.005, r = -0.40 (Table 4.5).

A similar trend was observed when withdrawal grades (W-grades) were analyzed (Table 4.5). A Mann-Whitney (U) test showed that the student withdrawal rate for the semesters that used the traditional laboratory curriculum (Median = 18.8%) were significantly higher than the withdrawal rate for the semesters that used the guided-inquiry laboratory curriculum (Median = 6.1%), U = 0.00, p = 0.05, r = -0.80. In an effort to support the decrease in withdrawal rate findings, additional data were collected and analyzed for all college-wide credit courses during the same semesters. The additional data produced a consistent college-wide withdrawal rate (Mean = 14.1%, SD = 1.3) over the six semester period.

Summary

The t-test analysis of students’ ACT scores by laboratory curriculum (traditional and guided-inquiry) determined no statistical difference between the two student test groups, thus both groups of students were treated as equivalent samples drawn from the same population.

The overall IRR value of (0.85, p < 0.01) indicated a fairly strong agreement with respect to the level of inquiry independently assigned to every laboratory experiment by each researcher. Accordingly, the final level of inquiry assigned to each laboratory experiment, and thus laboratory curricula overall, can be considered consistent and reliable. A statistical comparison of the inquiry levels for the two laboratory curricula indicated that the level of inquiry in the guided-inquiry curriculum (Median = Level-1) was significantly higher than that of the traditional curriculum (Median = Level-0). Furthermore, comparison of the modified for college instruction rubric used in this study and the definitions of inquiry levels (Table 2.0) by Pavelich
and Abraham (1979) further support the label of guided-inquiry as appropriate for the new laboratory curriculum.

The final analyses with respect to student performance and perseverance directly address both the research question, “How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at GRCC?” as well as the hypothesis, “The implementation of a more inquiry-based instructional laboratory curriculum will positively impact student performance and student perseverance in the laboratory portion of the first-semester general chemistry course (CM103) designed to educate students majoring in science and engineering programs.”

**Student Performance**

Comparison of the LRAFG by student group showed no significant differences. However, comparison of the ALRAFG by student group showed that the students (experimental group) in the guided-inquiry laboratory curriculum (median = 90.07%) scored significantly lower on their laboratory reports than the students (control group) in the traditional laboratory curriculum (median = 94.74%). Contrary to the hypothesis, adjusted laboratory report average final grades (ALRAFG) were found to be lower for the students enrolled in the guided-inquiry laboratory curriculum.

Statistical analysis revealed no significant difference between the laboratory report scores for Level-1 and Level-2 activities. However, laboratory report scores for Level-0 (Median = 9.50) activities were found to be significantly higher than the laboratory report scores for both Level-1 (median = 9.30) and Level-2 (median = 9.20) activities. Contrary to the hypothesis, the laboratory report scores were found to be lower for the guided-inquiry laboratory experiments.
Student Perseverance

Perhaps student perseverance produced the most compelling evidence as a result of the subgroup analysis of those students who submitted less than the 14 possible laboratory reports per semester. A statistical comparison of the subgroup of students revealed that the students in the inquiry-based laboratory curriculum submitted significantly more laboratory reports (Median = 12.00) than the students in the traditional laboratory curriculum (Median = 9.50). Withdrawal grades corroborate this finding as the withdrawal rate for the students in the inquiry-based laboratory curriculum (Median = 6.1%) was lower than that of the college-wide average for all courses (Mean = 14.1%) whereas the withdrawal rate for the students in the traditional laboratory curriculum (Median = 18.8%) was above that of the college-wide average for all courses (Mean = 14.1%). Thus, the traditional laboratory students showed a significantly higher withdrawal rate (Median = 18.8%) than the students in the inquiry-based laboratory (6.1%). However, when considering student withdrawal rates, the results must be interpreted with caution, particularly because there are many factors that influence this percentage (i.e. performance in the lecture component of the course, the economy, life experiences). With that said, these data suggest that the guided-inquiry laboratory curriculum had a positive effect on student perseverance in laboratory portion of the course, and perhaps the course as a whole.
Chapter Five: Conclusion

Summary of Study

The premise of this study was to use Fay et al.’s (2007) modified for college instruction rubric (Table 1.0/Table 3.0) to characterize the level of inquiry present in both the traditional and inquiry-based laboratory curricula, to examine the relationship between the level of inquiry present in a laboratory experience and student performance and perseverance outcomes, and to propagate a standard means of communication with respect to inquiry-based science instruction. As such, this investigation examined the research question, “How does the degree or level of inquiry-based science laboratory instruction impact student performance and student perseverance in the first-semester general chemistry course (CM103) at GRCC?” explored the hypothesis, “The implementation of a more inquiry-based instructional laboratory curriculum will positively impact student performance and student perseverance in the laboratory portion of the first-semester general chemistry course (CM103) designed to educate students majoring in science and engineering programs,” and documented the changes observed when a laboratory program was transitioned from a structured traditional laboratory curriculum to a more inquiry-based laboratory curriculum. Hence, this study informs and provides quantitative data and evidence to chemistry instructors, science educators, schools and institutions as they contemplate inquiry-based science curriculum reform.

Conclusion

Researchers used the modified for college instruction rubric (Table 1.0/Table 3.0) to determine the level of inquiry present in each laboratory experiment and developed a graphic distribution of experiments (Figure 4.0) with respect to levels of inquiry across laboratory curricula. The distribution showed that the traditional laboratory curriculum was (64.3%) Level-
0 and (35.7%) Level-1, while the guided-inquiry laboratory curriculum was (7.1%) Level-0, (71.4%) Level-1, and (21.4%) Level-2, respectively. A statistical comparison of the inquiry levels for the two laboratory curricula indicated that the level of inquiry in the guided-inquiry curriculum (Median = Level-1) was significantly higher than that of the traditional curriculum (Median = Level-0). Thus, this research established that the new guided-inquiry laboratory curriculum was more inquiry-focused in nature than the old traditional verification curriculum.

Statistical analysis, via a Mann-Whitney (U) test, of the students’ laboratory report average final grades (LRAFG) were determined to be the same for both curricula; however, the students’ adjusted laboratory report average final grades (ALRAFG), which were derived solely from those laboratory reports that were completed and submitted to the instructor for grading, and excluded any/all laboratory report scores of zero, were found to be statistically significant as the guided-inquiry curriculum (Median = 90.07%) yielded a lower percentage over the traditional curriculum (Median = 94.74%), U = 925.00, p < 0.005, r = -0.47.

Analysis, via a Mann-Whitney (U) test, of the individual laboratory report scores as a function of inquiry-level discovered that laboratory report scores from experiments rated Level-0 (Median = 9.50) were statistically higher than the laboratory report scores from experiments rated as Level-1 (Median = 9.30), U = 153487.00, p < 0.005, r = -0.25; and as Level-2 (Median = 9.20), U = 34933.50, p < 0.005, r = -0.25, respectively. These findings imply that the integration of even a moderate level of inquiry into a laboratory activity increases the responsibility of the students to learn and the amount of effort required of the students to construct knowledge, (as defined in Table 1.0 inquiry level descriptions), and as such, decreases students’ individual laboratory report scores by as much as 2-3%, thus resulting in lower adjusted laboratory report average final grades (ALRAFG).
Analysis, via Mann-Whitney (U) tests, of laboratory report completion rates were determined to be the same for both curricula. A statistical comparison of the subgroup, containing only those students who submitted less than the 14 possible laboratory reports for instructor grading, revealed that the students in the inquiry-based laboratory curriculum submitted significantly more laboratory reports (Median = 12.00) than the students in the traditional laboratory curriculum (Median = 9.50), U = 226.00, p < 0.005, r = -0.40. Withdrawal grades corroborate this finding as the withdrawal rate for the students in the inquiry-based laboratory curriculum (Median = 6.1%) was lower than that of the college-wide average for all credit courses (Mean = 14.1%) whereas the withdrawal rate for the students in the traditional laboratory curriculum (Median = 18.8%) was above that of the college-wide average for all credit courses (Mean = 14.1%). Thus, the traditional laboratory students showed a significantly higher withdrawal rate (Median = 18.8%) than the students in the inquiry-based laboratory (6.1%).

Therefore, the use of inquiry-based laboratory experiments places more demand and responsibility on the students. However, this demand is not so much as to discourage students, on the contrary, it appears the increased responsibility and demand required of the students to perform inquiry-based laboratory activities actually improved student perseverance in the laboratory portion of the course, and perhaps the course overall.

Discussion

These research findings with respect to student perseverance are similar to two studies discussed in the literature review (1) reported increased attendance retention (Cooper, 1994) and (2) reported a decrease in withdrawal rates (Farrell et al., 1999). However, those studies also found increased learning in inquiry-based lecture and laboratory courses via (1) increase in
lecture exams (Cooper, 1994) and (2) increase in overall course grades (Farrell et al., 1999). While this study cannot make any claims about the role of inquiry-based science instruction in the laboratory with respect to improved student learning, due to institutional grading limitations, these results indicate that students are more invested in the laboratory experience and will likely complete a course with an inquiry-based laboratory curriculum. Perhaps the use of inquiry-based laboratory instruction has the effect of making the laboratory experience more valuable for students, and as such increased their perseverance in the course. This appears to be the case even though the inquiry laboratories in this study were more demanding as measured by the students’ laboratory report scores.

**Recommendations**

Although student retention is certainly important, these findings also generate additional questions that require future investigation. First, how does this inquiry-based laboratory curriculum impact student learning in chemistry? Perhaps data with respect to student learning could be collected and analyzed via an assessment instrument such as the new American Chemical Society Exams Institute laboratory exam. Second, a supplemental student survey could gather information to further investigate why students elect to withdraw from or complete a course to potentially answer the question: Why were the withdrawal rates lower for the students enrolled in the inquiry-based laboratory curriculum?

**Dissemination**

Portions of this work and results of this study have been shared with GRCC in the form of an annual department report and the National Science Teachers Association by way of manuscript publication in the *Journal of College Science Teaching* (Scott & Pentecost, 2013). In addition, the National Science Teachers Association granted permissions to (1) include in thesis
the complete published research article (Appendix A), and (2) use, reproduce and include in this thesis any portion of the published research article (Appendix B).
References


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Appendix A-Published Research Article
From Verification to Guided Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?

By Pamela Scott and Thomas C. Pentecost

How does the degree of inquiry-based laboratory instruction impact student performance and student perseverance in the laboratory portion of a first-semester general chemistry course? The implementation of a new first-semester general chemistry laboratory curriculum provided an opportunity to address this question. A modified version of Lederman's continuum of scientific inquiry rubric (Fay, Grove, Towns, & Brez, 2007) was used to establish the degree of inquiry incorporated into each laboratory experiment in the old and new curricula. Laboratory report average final grades and individual laboratory report scores were used to measure student performance. A comparison of laboratory report completion between the two curricula was done to measure student perseverance. A final comparison sought to determine if any relationships exist between degree(s) of inquiry and student performance and perseverance.

Increasing the amount of inquiry-based instructional activities in both lecture and laboratory has been at the heart of most calls for curriculum reform at all levels (Bransford, Brown, & Cocking, 2000; Lederman, 2004). Analysis of the literature indicates that the effectiveness of inquiry-based laboratory instruction has been measured by changes in student attitudes in addition to quantitative comparisons of student learning. Further complicating the issue is the lack of a standard definition for an "inquiry" focused laboratory experience.

Pavelich and Abraham (1979) examined a guided/open-ended inquiry laboratory format in the general chemistry laboratory that relied on observational assessment of student learning improvement and attempted to measure growth in student mental maturity via a Piagetian-type test. The students in the guided/open-ended format had larger gains in intellectual development, and their description of the laboratory environment was in alignment with an inquiry environment. A direct comparison of inquiry and traditional laboratory environments (Hofstein, Nahum, & Shore, 2001; Kipnis & Hofstein, 2007) indicates that the inquiry format fosters the development of more positive student attitudes and perceptions of their learning. Students in the inquiry-based curriculum asked more questions, and these questions were more open-ended than those of the control group, suggesting that the teaching technique had a positive effect on encouraging inquiry. There is also evidence that the use of inquiry-based laboratory activities, both alone and with inquiry-based lecture instruction, increases student performance on lecture exams (Cooper, 1994; Cooper & Kerns, 2006; Greenbowe & Hand, 2005; Schroeder & Greenbowe, 2008). In a large-scale experiment Cooper (1994) found evidence that the use of cooperative learning with inquiry laboratories in general chemistry can impact student performance in lecture. In organic chemistry the effectiveness of inquiry is very dependent on the nature of the student groups and the interaction of the students and the instructor (Cooper & Kerns, 2006). These studies identified improvement in student learning that was due to inquiry in the lecture and laboratory; these studies did not evaluate the level of inquiry present in the laboratory experience.

It is often difficult to compare studies that investigate the effectiveness of inquiry-based teaching because the degree or level of inquiry used is often not characterized. The Continuum of Scientific Inquiry rubric (Lederman, 2004) was developed to characterize the level of inquiry for high school instruction and has been adopted for use in college settings by Fay, Grove,
Towns, and Bretz (2007). Rather than seeking to answer the question of whether inquiry-based instruction had value in laboratory application, the study focused on demonstrating that the level of inquiry present in a curriculum could be characterized. This work provided a rubric to aid instructors in selecting an inquiry-based curriculum before they mistakenly invest time implementing “new” programs that are in fact closer to a traditional verification curriculum.

The work reported in this article makes use of this rubric to characterize the level of inquiry of two laboratory curricula and to investigate the following question: How does the degree of inquiry-based laboratory instruction impact student performance and student perseverance in the laboratory portion of a first-semester general chemistry course?

**Methodology**

In 2008, a commercially published guided-inquiry laboratory curriculum (Kerner & Lamba, 2008) was instituted for the first-semester general chemistry course at a two-year community college. This change provided a case study of the new guided-inquiry laboratory curriculum versus the “old” traditional laboratory curriculum, which consisted of a collection of verification experiments written “in-house.” Researchers used a modified version of Lederman’s Continuum of Scientific Inquiry rubric (Fay et al., 2007) to assess both curricula and to establish the degree of inquiry incorporated into each laboratory experiment. Student performance was evaluated via laboratory report average final grades and individual laboratory report scores, whereas student perseverance was measured by comparing overall completion rates for laboratory reports by curriculum to determine if any relationships exist between degree(s) of inquiry and student performance and perseverance in the laboratory portion of the course. It is worth noting again that, because of institutional limitations, we are unable to link the changes in the laboratory format to improvement in student learning in chemistry. Without external, course objective–related measures of learning, we cannot make any claims about the impact on student learning. By using student scores on the laboratory reports as a measure of student performance, we hope to compare this aspect of student performance in the two curricula.

The data set consisted of students’ individual laboratory report scores and overall laboratory report final grades from intact laboratory sections, three semesters from the old traditional curriculum (fall 2006, winter 2007, and fall 2007) and three semesters from the new guided-inquiry curriculum (fall 2008, winter 2009, and fall 2009). In order to minimize instructor effect, data was collected from the same laboratory instructor for all six semesters. ACT test scores of the two groups of students were compared to establish population equivalency. There were a total of 183 students with allocated laboratory grades in the original data set. ACT scores were only available for 128 of these students. This smaller ($n = 128$) sample was used for statistical analysis. The control group consisted of 58 students who performed traditional verification laboratory experiments, whereas the experimental group consisted of 70 students who performed the guided-inquiry laboratory experiments.

The comparison of average ACT scores is shown in Table 1. The scores were normally distributed and have equivalent variances, both of which are requirements for comparison of the groups using a $t$-test. On average, there was no difference in the ACT comprehensive score of the students in the traditional curriculum and the guided-inquiry curriculum, $t(126) = -0.12, p > .05$. The same was true for the students’ ACT science scores, $t(126) = -2.12, p > .05$. Therefore, the two groups of students, those completing the old traditional verification laboratory activities and those completing the new guided-inquiry laboratory activities, were treated as equivalent samples drawn from the same population.

The level of inquiry present in the 14 laboratory experiments from each of the laboratory curricula (traditional and guided inquiry) was analyzed and characterized independently by two researchers using a modified version of the Continuum of Scientific Inquiry rubric (Fay et al., 2007), shown in Table 2. Laboratory experiments comprised of multiple parts were assigned a single inquiry level, which was determined by the highest level of inquiry present in any part.

All 28 laboratory experiments (14 from the old traditional verification and 14 from the new guided-inquiry curricula) were evaluated by both researchers independently using the rubric from Table 2, and then a consensus was developed. A Spearman rho (Field, 2005) correlation coefficient was used to assess the interrater reliability (IRR) of the level of inquiry classification for each laboratory curriculum as shown in Table 3. The closer the overall value is to 1, the stronger the relationship between the ratings of the two researchers. The overall IRR for the collection of experiments was 0.853 and is significant at $p < .01$.

Table 4 shows the final level of inquiry appraised for each experiment.
in both laboratory curricula. Note that for both curricula, the relevant chemistry concepts have been indicated in Table 4. When the level of inquiry for any given experiment resulted in disagreement, discussions between the two researchers negotiated a final level of inquiry. It's important to note that any initial differences regarding inquiry level were never more than one level different.

The distribution of the 14 laboratory experiments in each curriculum as rated across the levels of inquiry is shown in Figure 1. The majority of the experiments in the traditional laboratory curriculum were rated Level 0 (64.3%) with the remainder rated Level 1 (35.7%); conversely, the majority of the experiments in the guided-inquiry laboratory curriculum were rated Level 1 (71.4%) and Level 2 (21.4%), respectively.

A statistical comparison of the level of inquiry in the two curricula was done using a Mann-Whitney (U) test (Field, 2005), the nonparametric equivalent of a t-test. This more conservative comparison was warranted because the inquiry levels were not normally distributed within each curriculum. The level of inquiry in the guided-inquiry curriculum (Median = 1.00) was significantly higher than that of the traditional curriculum (Median = 0.00), U = 34.50, p < .005, r = -0.62. Likewise, comparison of the rubric used in this study and the definitions of inquiry levels by Pavelich and Abraham (1979) further support that the label of guided inquiry is appropriate for the new curriculum.

Before comparing student performance, it is important to note that the new inquiry laboratories were determined to be suitable replacements for the traditional experiments by the department. Therefore, these laboratories are not “watered-down” experiments and should have a similar level of expectations of the students.

### Results and discussion

#### Effect of curriculum on student performance

Initially, the student performance evaluation compared the laboratory report average final grade (LRAFG) of both student groups for both laboratory curricula. The LRAFG was calculated using all 14 laboratory report scores per semester and included any/all zero scores assessed for
laboratory reports that were not submitted for a score. Further evaluation compared an adjusted laboratory report average final grade (ALRAFG) of both student groups for both laboratory curricula. The ALRAFG was derived solely from those laboratory reports that were completed and submitted for a score, thus it excluded any/all laboratory report scores of zero. Table 5 shows both comparisons by student group.

The LRAFG and ALRAFG for both student groups were statistically compared using the Mann–Whitney test because they were not normally distributed. Data analysis revealed that the LRAFG for students in the guided-inquiry curriculum (Median = 85.00) is not significantly lower than that of students in the traditional curriculum (Median = 89.93). On the other hand, the ALRAFG for students in the guided-inquiry curriculum (Median = 90.07) is significantly lower than that of students in the traditional curriculum (Median = 94.74), U = 925.00, p < .005, r = −0.47. Given that both types of labs were taught and graded by the same person and covered similar topics, these data may indicate that expectations of students in the guided inquiry curriculum are higher and that the inquiry reports might be better probes of student understanding than traditional lab reports.

Effect of level of inquiry on student performance

The second student performance evaluation considered all students as one population and compared individual laboratory report scores by the level of inquiry. For ease of comparison, the scores for each laboratory report were converted to a 10-point scale. Ideally, had every student submitted each laboratory

<table>
<thead>
<tr>
<th>Table 4</th>
</tr>
</thead>
<tbody>
<tr>
<td><strong>Laboratory curricula: Level of inquiry.</strong></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Experiment: Traditional laboratory curriculum</th>
<th>Inquiry level</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Introduction to the Chemistry Lab</td>
<td>0</td>
</tr>
<tr>
<td>2. Determination of Density</td>
<td>0</td>
</tr>
<tr>
<td>3. Separation of a Mixture (extraction using acid base properties)</td>
<td>0</td>
</tr>
<tr>
<td>4. Electrons &amp; Light (atomic spectra and unknown identification)</td>
<td>1</td>
</tr>
<tr>
<td>5. Chemicals &amp; Reactions (observe, describe, and identify chemical changes)</td>
<td>1</td>
</tr>
<tr>
<td>6. Solution Chemistry (determining the concentration of a solution)</td>
<td>0</td>
</tr>
<tr>
<td>7. Molecular Modeling (Lewis dot structures and geometries)</td>
<td>0</td>
</tr>
<tr>
<td>8. Separation of Compounds (intermolecular forces)</td>
<td>1</td>
</tr>
<tr>
<td>9. A Cycle of Copper Reactions (reaction types)</td>
<td>0</td>
</tr>
<tr>
<td>10. Stoichiometry</td>
<td>0</td>
</tr>
<tr>
<td>11. Determining Concentration (Beer's law)</td>
<td>0</td>
</tr>
<tr>
<td>12. Enthalpy Changes &amp; Hess's Law (calorimetry—heat of solution)</td>
<td>0</td>
</tr>
<tr>
<td>13. Signs of △H, △S &amp; △G (qualitative thermodynamics)</td>
<td>1</td>
</tr>
<tr>
<td>14. Ten Unknowns (experimental design and unknown identification)</td>
<td>1</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Experiment: Guided-inquiry laboratory curriculum</th>
<th>Level of inquiry</th>
</tr>
</thead>
<tbody>
<tr>
<td>1. Introduction to the Chemistry Lab</td>
<td>0</td>
</tr>
<tr>
<td>2. How Long Can a Bubble Last? (experimental design)</td>
<td>2</td>
</tr>
<tr>
<td>3. Are All Pennies the Same? (density)</td>
<td>1</td>
</tr>
<tr>
<td>4. What Makes a Solution Colored or Colorless? (absorbance, color, &amp; periodic trends)</td>
<td>1</td>
</tr>
<tr>
<td>5. What Factors Affect Color Intensity? (solution prep &amp; Beer's Law)</td>
<td>1</td>
</tr>
<tr>
<td>6. Are There Property Patterns? (reactivity and periodic properties)</td>
<td>1</td>
</tr>
<tr>
<td>7. What Factors Affect the Solubility of Ions? (double replacement reactions)</td>
<td>2</td>
</tr>
<tr>
<td>8. Can Toxic Ions Be Removed From Water by Precipitation? (separation of compounds by solubility)</td>
<td>2</td>
</tr>
<tr>
<td>9. Do Like Repel or Attract? (intermolecular forces and solubility)</td>
<td>1</td>
</tr>
<tr>
<td>10. How Much Hydrogen? (stoichiometry &amp; reactivity of metals with acid)</td>
<td>1</td>
</tr>
<tr>
<td>11. How Much Is Too Much? (stoichiometry &amp; limiting reagent)</td>
<td>1</td>
</tr>
<tr>
<td>12. Which Salts Make Good Cold Packs &amp; Hot Packs? (calorimetry—heat of solution)</td>
<td>1</td>
</tr>
<tr>
<td>13. How Is Heat Measured Indirectly? (calorimetry—Hess's law)</td>
<td>1</td>
</tr>
<tr>
<td>14. Thermodynamic Signs (qualitative thermodynamics)</td>
<td>1</td>
</tr>
</tbody>
</table>

Note: Bold italicized text denotes initial disagreement, and the value posted is the agreed-on level.
report for a score, there would have been 1,792 individual laboratory report scores. However, scores were only available for 1,564 laboratory reports; consequently the comparison was made using these and is shown in Table 6. These data were not normally distributed so the comparison was done using the Kruskal–Wallis (H) test, nonparametric equivalent of an analysis of variance (Field, 2005), resulting in a significant effect with respect to level of inquiry on the individual laboratory report scores, H(2) = 92.34, p < .005. Mann–Whitney tests with a Bonferroni correction were used to determine which level of inquiry was different from the others. Statistical analysis determined that the laboratory report scores for activities rated as Level 1 (Median = 9.30) or Level 2 (Median = 9.20) were not significantly different. Alternatively, laboratory report scores for activities rated as Level 0 (Median = 9.50) were statistically higher than those scores for activities rated as Level 1 (Median = 9.30), U = 153,487.00, p < .005, r = -0.25, and Level 2 (Median = 9.20), U = 34,933.50, p < .005, r = -0.25, respectively.

The data suggests that the incorporation of inquiry, at any level, into laboratory activities increases the level of difficulty of the activity, which in turn yields lower laboratory report scores and lower laboratory final grades. This is a reasonable finding because one goal of inquiry instruction is to enhance the development of critical-thinking skills, a process that requires effort.

Level of difficulty and student perseverance
To see if the increased efforts required of students to learn knowledge through guided-inquiry activities affected student perseverance, overall completion rates for laboratory reports by curriculum were compared. The traditional laboratory curriculum produced 672 completed laboratory reports out of a possible 812, resulting in a completion rate of 82.76%. Additionally, the guided-inquiry curriculum yielded 892 completed laboratory reports out of a possible 980, which resulted in a completion rate of 91.02%. Although these results appear different, a Mann–Whitney test indicates that the results are not statistically different. A subgroup statistical analysis of student perseverance for students who did not complete all 14 laboratory reports was used to capture the effect of curriculum on students who might be in danger of not completing the course. This analysis revealed that students in the guided-inquiry curriculum completed more laboratory reports (Median = 12.00), and their completion rate is significantly higher than that of students in the traditional curriculum (Median = 9.50), U = 226.00, p < .005, r = -0.40.

A similar trend was observed when withdrawal grades (W-grades) were analyzed. A Mann–Whitney test shows the student withdrawal percentage for the semesters that used the traditional laboratory curriculum (Median = 18.8) was significantly higher than the withdrawal percentage during the semesters during which the inquiry laboratory curriculum was used (Median = 6.1), U = 0.00, p = .05, r = -0.80. In an effort to support the decrease in withdrawal percentage findings, additional data was collected and analyzed for all collegewide credit courses during the same semesters. The additional data produced a consistent collegewide withdrawal percentage (M = 14.1, SD = 1.3) over the six-semester period.

When considering student withdrawals, the results must be interpreted with caution, particularly because there are many factors that influence this percentage (i.e., performance in the lecture component of the course, economy, etc.). With that said, these data suggest that the guided-inquiry laboratory curriculum had a positive effect on student perseverance in the laboratory portion of the course and perhaps the course as a whole.

Summary
This study shows that the new guided-inquiry laboratory curriculum was more inquiry focused in nature than the old traditional verification curriculum. The students' LRAFGs were statistically the same for both curri-
Although student retention is certainly important, these findings also generate additional questions that we would like to try and investigate in the future. First, how does this inquiry-based curriculum impact student learning in chemistry? We envision being able to collect student learning data with an assessment such as the new American Chemical Society Exams Institute laboratory exam.

Second, we would like to investigate why students choose to drop out of or stay in a course to further answer the question: Why were the withdrawal rates lower with the inquiry-based curriculum?

**Acknowledgments**

We thank Bill Faber, the general chemistry laboratory instructor who enabled this research to be conducted and Sherril Soman, Ellen Yezierski, Deborah Herrington, Julie Hende methane, and the Target Inquiry Cohort for their guidance and support. This paper is based in part on research and professional development supported by the National Science Foundation (ESI-0553215) and the Target Inquiry program at Grand Valley State University. Any opinions, findings, and conclusions or recommendations expressed in this article are those of the authors and do not necessarily reflect the views of the National Science Foundation or Grand Valley State University.

**References**


Fay, M. E., Grove, N. P., Towns, M. H., 

**TABLE 5**

Comparisons of LRAFG and ALRAFG by student group.

<table>
<thead>
<tr>
<th>Student group (lab curriculum)</th>
<th>Average/median % LRAFG (SD)</th>
<th>Average/median % ALRAFG (SD)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Control, N = 58 (traditional)</td>
<td>89.93/77.59 (27.25)</td>
<td>94.74/93.89 (3.83)</td>
</tr>
<tr>
<td>Experimental, N = 70 (guided inquiry)</td>
<td>85.00/81.00 (15.29)</td>
<td>90.07/88.75 (7.01)</td>
</tr>
</tbody>
</table>

Note: LRAFG = laboratory report average final grade; ALRAFG = adjusted laboratory report average final grade.

**TABLE 6**

Comparison of individual laboratory report score by level of inquiry.

<table>
<thead>
<tr>
<th>Level of inquiry</th>
<th>Number of laboratory reports</th>
<th>Median/average laboratory report score out of 10 possible points (SD)</th>
</tr>
</thead>
<tbody>
<tr>
<td>0</td>
<td>509</td>
<td>9.50/9.39 (0.84)</td>
</tr>
<tr>
<td>1</td>
<td>855</td>
<td>9.30/8.99 (1.04)</td>
</tr>
<tr>
<td>2</td>
<td>200</td>
<td>9.20/8.86 (1.27)</td>
</tr>
</tbody>
</table>


Pamela Scott is a laboratory coordinator in the Physical Science Department at Grand Rapids Community College in Grand Rapids, Michigan. Thomas C. Pentecost (pentecot@gvsu.edu) is an assistant professor in the Department of Chemistry at Grand Valley State University in Allendale, Michigan.
Appendix B-Copyright Permission Forms
Hi Pam, you have our permission. Best of luck!

Dear National Science Teachers Association,

I'm contacting you to request permission to use a published article as my thesis project for a Master's of Education degree. My name is Pamela Scott and my manuscript *From Verification to Guided Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?* was published in the *Journal of College Science Teaching*, January/February 2013, Vol. 42, No.3.

Thank you,

Pam Scott

Physical Science Laboratory Coordinator
Grand Rapids Community College
143 Bostwick NE
Grand Rapids, MI 49503
(616) 234-4297 office
(616) 234-3590 fax
May 8, 2014

Emily Brady: ebrady@nsta.org

Dear Ms. Brady,

I am currently enrolled in the Grand Valley State University (GVSU) Graduate Studies in Education Program, and I am writing a thesis for the completion of my Master’s in Education. My thesis is entitled “From Verification to Guided Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?” May I receive permission to include in the appendices a copy of the published article?

Your signature at the bottom portion of this letter confirms your ownership of the above item. The inclusion of your copyrighted material will not restrict your re-publication of the material in any other form. Please advise if you wish a specific copyright notice to be included on each page. My thesis will be cataloged in the GVSU library and will be available to other students and colleges for circulation.

Sincerely,

Pam Scott
Grand Rapids Community College
143 Bostwick NE
Grand Rapids, MI 49329
Phone: (616) 234-4297
Fax: (616) 234-3590
E-mail: pscott@grrc.edu

PERMISSION IS GRANTED to you (Pam Scott) to include the requested material(s) in her GVSU Master’s of Education thesis.

Name of Company/Organization

Permission granted by: Emily Brady, NSTA

Title: Manager permissions

Date: 5/9/14

69
June 28, 2014

info@youthlearn.org

Dear youthlearn.org:

I am currently enrolled in the Grand Valley State University (GVSU) Graduate Studies in Education Program, and I am writing a thesis for the completion of my Master’s in Education. My thesis is entitled “From Verification to Guided-Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?” May I receive permission to include in my thesis and appendices a copy of the following item?


Your signature at the bottom portion of this letter confirms your ownership of the above item. The inclusion of your copyrighted material will not restrict your re-publication of the material in any other form. Please advise if you wish a specific copyright notice to be included on each page. My thesis will be cataloged in the GVSU library and will be available to other students and colleges for circulation.

Sincerely,

Pam Scott
Grand Rapids Community College
143Bostwick NE
Grand Rapids, MI 49503
Phone: (616) 234-42978
Fax: (616) 234-3590
E-mail: pscott@grcc.edu

PERMISSION IS GRANTED to you (Pam Scott) to include the requested material(s) in (his or her) GVSU Master’s of Education thesis.

Name: [Signature]

Address: Education Development Center Inc., 43 Foundry Ave, Waltham, MA 02453

Date: July 3, 2014

NOTE: Please give credit for this image to “The YouthLearn Initiative at Education Development Center, Inc. © 2001”
July 7, 2014

RE: “Inquiry Process Graphic” (Hereinafter, the “Work”) and Pam Scott (the “Requestor”).

Dear Pam Scott:

Education Development Center, Inc. (“EDC”) is pleased to grant permission to the Requestor to use the Work in the thesis entitled “From Verification to Guided-Inquiry: What Happens When a Chemistry Laboratory Curriculum Changes?” The fee for this permission is waived for the limited, non-commercial purposes as described in this request. The requested materials may not be re-used without the written permission of EDC. EDC shall retain copyright to the requested work.

To accept this permission, please arrange for an authorized representative from the Requestor to accept and execute this permission in the space provided below and return a fully executed copy of this letter to me.

Please use the following attribute/credit line on all copies of the Work:

Copyright 2014 Education Development Center, Inc.  
Reprinted with permission with all other rights reserved. 

Sincerely,

Christine Filosa  
Director and Senior Attorney, Office of Legal Affairs

Agreed and Approved

Pam Scott

Signature

Pam Scott

Date

7/10/14

Name

Graduate Student

Title
Appendix C-Institutional Review Board (GRCC) Forms
September 22, 2009

Ms. Pam Scott
Faculty – Physical Science
Grand Rapids Community College
143 Bostwick Ave NE
Grand Rapids, MI 49503

Dear Ms. Scott:

TITLE OF PROPOSAL: Target Inquiry: How Do Students Respond to Inquiry Instruction?

This letter is to officially notify you of the approval of your project by the Institutional Review Board (IRB) at Grand Rapids Community College. It is the Board's opinion that you have provided adequate safeguards for the rights and welfare of the participants in this study. Your proposal has been classified as "Exempt."

You are responsible for immediately informing the Institutional Review Board of any changes to your protocol, or of any previously unforeseen risks to the research participants.

This approval is good from September 1, 2009 to September 1, 2010. If you wish to continue your research after this date, you must complete and submit an updated protocol.

Please let me know if you have any questions.

Sincerely,

Donna Kragt
Dean of Institutional Research and Planning
Chair of the IRB
Grand Rapids Community College
INSTITUTIONAL REVIEW BOARD (IRB)
APPLICATION FOR APPROVAL TO USE HUMAN SUBJECTS IN RESEARCH

1. Project Title: Target Inquiry: How do Students Respond to Inquiry Instruction?

PRINCIPAL INVESTIGATOR INFORMATION

2. Principal Investigator: Pam Scott
   Department: Physical Science  Phone: 234-4297  Fax: 234-3590
   Email: pscott@grcc.edu  *(Required)*

3. Co-PI (if any): Dr. Deborah Herrington & Dr. Ellen Yezierski
   Department: GVSU Chemistry  Phone: 616-331-3317  Fax: 616-331-3230
   Email: herringd@gvsu.edu & yezierse@gvsu.edu  *(Required)*

4. Status (check one): □ Faculty □ Student  X Other (please explain): Pam is faculty @ GRCC, Deborah & Ellen are faculty at GVSU

   For student and non-GRCC researchers only, please give your home address and phone number:
   Dr. Herrington: 1867 Sun Park Drive, Zeeland MI 49464, (616)741-9124
   Dr. Yezierski: 14903 Hawthorne Street, Grand Haven MI 49417, (616)-850-8411

PROTOCOL INFORMATION)*

5. Does your study involve individually identifiable protected health or mental health information (PHI), including demographic information and biological specimens identified to an individual, created or maintained by, or received from, a person or an entity covered by the Privacy Rule issued under the Health Insurance Portability and Accountability Act (HIPAA) (e.g., a hospital; a physician, or a practice in psychology, psychotherapy, or social work; a health insurer, HMO, or health plan; or a community clinic, or a social service or mental health agency)?  ___Yes  X No

6. If your answer to question (5) is Yes, please list below or on a separate sheet the PHI that is necessary for your research and that you intend to use in your research.

   _____________________________________________________________

7. If your answer to question (5) is Yes, please list below or on a separate sheet the name and address of each person or entity that is creating, maintaining or providing the PHI for your research.

   _____________________________________________________________

   _____________________________________________________________
8. Does your study involve the collection of data from a vulnerable population? If yes, please specify type of population:

For a complete list of categories of vulnerable populations, as well as the special safeguards required when conducting research with them, see pages 9-10 of the PI manual. Special Informed Consent procedures are necessary when conducting research with minors. See page 20 of the PI Manual for information.

☐ Yes  X No
☐ Children/Minors ☐ Prisoners ☐ Fetuses ☐ Pregnant Women ☐ Cognitively Impaired Persons ☐ Other _______

9. Does this study involve deception (research in which the subject is purposely led to have false beliefs or assumptions)?

☐ Yes  X No

10. If the study involves risk to subjects, is the risk greater than that incurred in ordinary life or tasks?

☐ Yes  X No

11. Has this study ever been previously approved by this IRB?

☐ Yes  X No

12. Is this proposal new or revised in response to previous IRB review?

X New  ☐ Revised

☐ Yes  X No

13. Is funding being sought for this study? If yes, through what sponsoring agency?

Agency: ____________________________________________________________

I certify that the research plan and safeguards to human subjects described in this application conform to that which has been submitted/will be submitted to an external funding source.

Principal Investigator: _______________________________________________

Date: _____________________________________________________________

14. Is this study being reviewed by an IRB at another institution? If yes, please list the institutions below.

☐ Yes  X No

Grand Valley State University (GVSU)

Documentation of IRB reviews of this study conducted at other institutions must be provided when it becomes available. Research may not begin until IRB review has been concluded at all institutions involved.
Please answer the following questions on a separate sheet.

15. State the purpose of the research. Include major hypotheses and research design. If the study is part of a larger study, briefly describe that larger study and indicate whether it has received IRB approval from another institution. Please keep in mind that the IRB is composed of individuals from many disciplines and thus the description of your research should be written in terms readily comprehensible by non-experts.

This study is part of the Target Inquiry (TI) program at GVSU designed to increase the frequency and quality of inquiry instruction in the classroom. As part of this program Pam Scott was involved in designing new inquiry activities and is now looking to use action research to study the impact of inquiry instruction on students. The action research project portion of the TI program has been approved by GVSU’s IRB. The IRB application, renewal, and approval letters can be found in the appendix.

GRCC’s Physical Sciences Department recently instituted a guided-inquiry lab curriculum for CM103 and CM104, the general chemistry course sequence. The change in laboratory curriculum provides an opportunity to study the impact of the "new" guided-inquiry lab curriculum compared to the "old" traditional lab curriculum on student success. Lederman’s Continuum of Scientific Inquiry rubric will be used to assess both curricula to establish the degree of inquiry incorporated into each lab experiment. Student performance will then be evaluated using individual laboratory report grades as well as final course grades to determine if any correlations exist between degree of inquiry and student performance. Additionally, we will collect data concerning course retention/attendance (lab and lecture) in general chemistry to determine whether or not the laboratory curriculum impacts these factors. Laboratory and final course grades for CM103 will be collected from participating faculty members, and data collection will tentatively be completed by May 2010.

16. Describe the source(s) of subjects and the selection criteria. Selection of subjects must be equitable and, in the case of protected populations such as children, prisoners, pregnant women, the mentally disabled, etc. should address their special needs. Include the number of subjects. The text of any advertisement, letter, flier, oral script or brochure used to solicit potential subjects must be attached.

Physical Sciences faculty member, Mr. Bill Faber, has agreed to provide the student data (individual lab report grades and course attendance/retention) for this study.

- CM103 "old" traditional lab curriculum:
  - Fall 2006 1 section 24 students
  - Winter 2007 2 sections 48 students
  - Fall 2007 1 section 24 students

- 96 students (max.)

- CM103 "new" guided-inquiry lab curriculum:
  - Fall 2008 1 section 24 students
  - Winter 2009 2 sections 48 students
  - Fall 2009 1 section 24 students

- 96 students (max.)

17. Provide a description of the procedures to be followed. If available, include copies of questionnaires and/or interview protocol, or a sufficiently detailed description of the measures to allow the IRB to understand the nature of subjects’ involvement.

As this project involves comparing student scores for the sections that did not use the inquiry lab curriculum and the section(s) that did use the inquiry curriculum in Mr. Faber’s sections of CM103, the participants will not be directly involved with the researcher in any way. Student ACT scores, final course grades and identification of lecture instructor, will be obtained from GRCC’s Institutional Research and Planning Department for all students enrolled in the "old" and "new" courses to help establish the equivalency of the student groups. Individual student laboratory report grades will be provided by Mr. Faber. As all findings will be reported using class aggregate data, student names will not be attached to the scores in any way. Additionally, in recording the data each student will be assigned a unique, unidentifiable number so that the data cannot be linked to the student.
18. Describe any potential harms or benefits to be derived by subjects, with a discussion of the risk/benefit ratio. For approval of any study with more than minimal risk, the benefits must clearly be shown to outweigh the risk. Describe how the study may expose participants to stress, physical, psychological or interpersonal hazard, including the possibility of pain, injury, disease, discomfort, embarrassment, worry or anxiety.

As these data are readily available, require no interaction with the participants, and cannot be linked to a particular participant with the use of a unique identifier number, there are no potential harms to the participants. Although there are no benefits to the current participants, analysis of the impacts of the new inquiry laboratory curriculum on student achievement in the CM103 course can help inform future curricular changes and therefore there is a potential benefit to future students.

19. Describe the specific methods by which confidentiality and anonymity will be protected, including the use of data coding systems, how and where data will be stored and who will have access to it, and what will happen to data after the study has been completed.

See attached Adult Consent Form.

20. If applicable, provide the following: 1) a description of the debriefing procedures to be used in cases where deception has occurred; 2) a statement describing what actions you will take should the research reveal the possibility of a medical or other potentially troubling condition.

Not Applicable.

21. Describe the oral and written consent processes and attach all consent documents. When the consent form to be used will be in a language other than English, an English translation must be provided. Unless one or more of the required elements described below is explicitly waived by the IRB, informed consent documents should contain:
   A. A fair explanation of the purposes of the research and the expected duration of the subject's participation, a description of the procedures to be followed, and identification of any procedures which are experimental;
   B. A description of any possible discomforts and risks reasonably expected. This includes any potential financial risks that could ensue;
   C. A description of any benefits reasonably expected;
   D. A disclosure of any appropriate alternative procedures;
   E. A statement that participation is voluntary, refusal to participate will involve no penalty or loss of benefits to which the subject is otherwise entitled, and the subject may discontinue participation at any time without penalty or loss of benefits to which the subject is otherwise entitled;
   F. An offer to answer any inquiries concerning the goals of the research or the research procedures and to provide a summary of results upon request and an explanation of whom to contact for answers to pertinent questions about the research and research subjects' rights, and whom to contact in the event of a research-related injury to the subject;
   G. An instruction that the subject is free to withdraw or discontinue participation at any time without prejudice.
   H. A statement describing the extent, if any, to which confidentiality of records identifying the subject will be maintained; and
   I. Provisions for parent or guardian approval for participation of minors or for subjects from vulnerable populations when appropriate.

Upon approval of the study, the consent document will be stamped with an expiration date. Only this document may be used when enrolling subjects. Studies extending beyond the expiration date must be submitted for a continuation review. Any changes in the consent form must be approved by the IRB.

This research is considered exempt under 45 CRF 46.101(b)(1) (research in an established educational setting comparing instructional techniques), The students are not asked to do anything outside of their normal classroom activities, and by using class aggregate data information cannot be linked to a particular student.

22. Please provide any other information that might be pertinent to the IRB's decision.
Appendix D-Human Research Review Committee (GVSU) Forms
Principal Investigator(s): Deborah Herrington and Ellen Yezierski
Contact email address: herringd@gvsu.edu and yezierse@gvsu.edu
Address and Telephone
Number of Principal Investigator(s): Deborah Herrington: 373 PAD 331-3809; Ellen Yezierski: 368 PAD 331-3808
GVSU Department or School: Chemistry Department
Title of the Project: Target Inquiry: How do Students Respond to Inquiry Instruction?

Date(s) and Location(s) of Subject Enrollment: College and high school instructor participants will be enrolled in the study spring and summer 2007. Student participants will be enrolled in the study beginning in September, 2007. Voluntary student participants will be recruited from the high school chemistry classes of the teacher researchers at the following area high schools: Allendale High School, Holland High School, West Ottawa High School, Black River Public School, North Muskegon High School, Muskegon High School, Western Michigan Christian High School, Kelloggsville High School, Jenison High School, and Hudsonville High School.

Summary of the Project: 'see attached' is not acceptable
The teacher researchers for this project will be involved in the development teaching materials that appropriately model the process of scientific inquiry in their classrooms. During the 2007-2008 school year, the teachers will implement their new materials in their classrooms and use action research to evaluate the impact of these materials on their students. The research questions that will guide the teacher researchers’ evaluation of their materials are:

1. How do inquiry activities impact students’ conceptual understanding of chemistry?
2. How do inquiry activities impact students’ science processing skills?
3. How do inquiry activities impact students’ attitudes towards chemistry?

In what capacity does this project involve human subject? (E.g., surveys, interviews, clinical trial, use of medical records, etc.)
Participants will complete surveys and content tests linked to the teachers’ curriculum materials. Course materials such as lab notebooks, test or quiz answers, homework problems, group activities, and projects may also be collected. Many of the participants in this study will be minors; therefore, we request an expedited review as described in 46.110 of the Federal Register under research category (7) “Research on individual or group characteristics or behavior.”

Check one:
_____This is a request for exemption from HRRC approval requirements as specified by 46.101 of the Federal Register 4616:8336, January 26, 1981. (Refer to instructions on the reverse of this form.)
X This is a request for expedited review as described in 46.110 of the Federal Register 46(16):8336, January 26, 1981. (Refer to instructions on the reverse of this form.)
_____This is a request for full review. (Refer to instructions on the reverse of this form.)
A. **Investigators**

Deborah Herrington: Grand Valley State University Department of Chemistry  
Ellen Yezierski: Grand Valley State University Department of Chemistry  
Brian Brethauer: Allendale High School  
Kevin Conkel: Hudsonville High School  
Tim Ewald: Black River Public School  
Deborah Johnson: North Muskegon High School  
Alice Putti: Jenison High School  
Gretchen Ludeman: Kelloggsville High School  
Peter Larsen: Holland High School  
Susan Munster: Muskegon High School  
Brian Vanzanten: West Ottawa High School  
Sarah Toman: Western Michigan Christian High School

B. **Location**

The inquiry materials will be developed at Grand Valley State University during Summer 2007. Teacher researchers may solicit information from local high school and college instructors to assist in the development of their inquiry materials. The evaluation data including chemistry content tests, surveys, and coursework materials will be collected at the 10 area high schools previously specified. Permission will be obtained from each of the high school principals and the parents prior to any data collection. Student assent will also be obtained. (A copy of the principal permission letter, parent consent, and student assent letters are in Appendix A.)

C. **Methods**

During spring and summer 2007, local high school and college instructors will be sent a voluntary survey to ascertain the chemistry content and process skill expectations for students entering college level chemistry courses. As these surveys will be anonymous, completion of the survey will imply participant consent. Data from these surveys will assist in the development of the inquiry materials. At the beginning of the 2007-2008 school year, informed consent will be obtained from the parents and assent from the students for the use of classroom content tests, survey, and coursework materials to evaluate the impact of new inquiry instructional materials on students’ conceptual understanding of chemistry, science processing skills, and attitudes towards chemistry. The goal of collecting this data is to allow teachers to further improve their instructional materials for themselves as well as other teachers who may wish to use their materials. Parents and students will be assured that any data obtained through tests, surveys, and course materials will be kept strictly confidential. Teachers will distribute the parent consent and child assent letters and will oversee their collection. To ensure confidentiality, all data from tests, surveys, and course materials will be viewed only by the investigators and the individual participant. Names will be removed from any of the materials and a code number will be used to track each participant’s data. Any materials used for publication will either be aggregate data from a class or use a pseudonym to protect the identity of the participants. All paper records will be kept in a locked filing cabinet in the teachers’ locked offices during the academic year to allow the teachers access for data analysis purposes. At the conclusion of the academic year, paper records will be stored in a locked file cabinet in a locked GVSU office (PAD 373 or PAD 368). Any computer data will be stored in password protected computer files. The records will be kept for a period of 3 years following the study to allow for completion of the evaluation and then destroyed. All tests,
surveys, and course assignments that are part of the standard course work will be required of all students; however, data for analysis will not be included for any student whose parent does not want them involved with the study.

Teachers will be videotaped up to 4 times per academic year during lessons that they identify as inquiry based and invite us to observe. At this time, any student who has not returned a signed permission form will be situated in the room so that image is not captured on tape. At the beginning of the class the person videotaping will remind the students that if at any time they wish to have videotaping terminated it will in no way influence their grade or relationship with their teacher. The videotapes will not be released or published and will only be viewed by the researchers, their undergraduate or graduate student working on the TI project, and the classroom teacher that was videotaped. The PIs and their graduate or undergraduate student will code each of the videotapes using the Reformed Teaching Observation Protocol\(^1\). At the end of the TI program the teachers will be asked to watch their classroom videotapes to reflect on the development of their teachers over the course of TI. The videotapes will be kept in a locked filing cabinet in PAD 373 or PAD 368 for 3 year following completion of the TI program to allow for data analysis. They will then be erased.

D. Potential Risks and Benefits

There are no risks to students participating in this study. The majority of data collected from the students will be standard course work. Additional surveys or content tests may help students think differently about the process of science or provide them with additional practice in taking standardized tests. There are several expected benefits from students engaging in the new inquiry instructional activities.

(1) Students will experience a more authentic science experience.
(2) Students may have their misconceptions challenged and as such develop the correct scientific explanations for phenomena.
(3) Students may gain a deeper understanding of key chemistry concepts.
(4) Students may gain a more accurate idea about the process of science.
(5) Students may improve their ability to think scientifically and critically.
(6) Students may improve their problem solving and data analysis skills.

The videotapes will in no way affect a student’s success in the course or their relationship with their teachers. The teachers will not view the videotapes until after their participation in TI has concluded. Although there are not any direct benefits to the students from being videotaped, the videotapes have potential benefits for teachers and their future classes.

(1) Teachers’ classroom practices will be documented over a 3-5 year period allowing them to critically reflect on and improve their teaching.
(2) Teachers will be able to identify strengths and weaknesses in their teaching that will allow them to better facilitate activities for future classes.

E. Drug or Devices to be Used

No drugs or devices will be used on TI teachers.

F. Granting Agencies

The previously mentioned 10 area high schools have each been $500 to support the teachers’ implementation of the new inquiry instructional activities. This funding has been provided by the National Science Foundation.

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Human Research Review Committee  
Change in Protocol Form

Date: 05/18/09

Principal Investigator(s): Deborah Herrington and Ellen Yezierski

Contact email address: herringd@gvsu.edu

Title of the Research Protocol: Target Inquiry: How do Students Respond to Inquiry Instruction?

Protocol File number: 07-243-H

Directions:
1. Explain only but exactly the change(s) that you want to make in the approved protocol.
   
   We are adding new teacher researchers to the project senior personnel. The new teachers researchers and their schools are as follows:
   
   Pam Scott Grand Rapids Community College
   Dale Eizenga Holland Christian
   Chad Bridle Grandville High School
   Deanna Cullen Whitehall High School
   Angela Slater Muskegon Heights High School
   Joseph O'Malley City High School
   Michelle Mason Portage Northern High School
   James Doug Mandrick Portage Central High School

2. Justify the change(s) and explain why it could not reasonably have been anticipated and incorporated at the time the original application for protocol review was submitted.
   
   Target Inquiry is a cohort program that enrolls new cohort of teachers every two years. The first cohort of teachers completed the program and is no longer with the program. The teacher researchers listed above are the second cohort of TI teachers, and they began the program in January of 2008, after the initial approval for the project was obtained. The teachers in the second cohort will be conducting action research projects related to the inquiry labs they develop during summer 2009, like those that the first cohort of teachers conducted. Although the specific questions that these teachers choose to investigate may be slightly different, the types of data they collect and the data collection and storage methods will be the same.

3. Explain whether in your judgment the proposed change materially effects (a) the level of risk to the participants (YES) X (NO), or (b) the relationship of the benefits to the risks of participation in the study as a whole (YES X (NO).

4. In your judgment does the change(s) warrant either or both of the following?
   (a) re-consenting the previously enrolled participants? (YES) X (NO)
   (b) a new consent form? (YES) X (NO) If YES, attach a new consent form and clearly indicate the changes therein.

5. You can add a new group of study participants to be recruited by following the change in protocol procedure. Each step needs to be addressed separately. Note that new invitation to participate letters, a description of the distribution method for that letter, method for selection of subjects (if more express an interest in participating than can be accommodated), and a new consent form need to be submitted for review.
Appendix D

Informed Consent Letters
Dear Parent,

Your child's chemistry teacher is a part of the Target Inquiry program at Grand Valley State University (GVSU). This program is designed to help teachers increase the quantity and quality of inquiry instruction in their chemistry classrooms. Research has shown that inquiry instruction can help students learn and retain chemistry concepts more effectively. We are conducting a study to determine how a new teacher professional development program affects teachers and student achievement in chemistry.

We are requesting your child's participation. The study will take place in your child's classroom and require 2 hours during the academic year during the regularly scheduled school day. Your child's participation in this study is voluntary. You (or your child) are free to decide not to participate in this study or to withdraw at any time without adversely affecting your relationship with you teacher, the investigators, or GVSU. Your decision will not result in any loss of benefits to which your child is otherwise entitled. Specifically, your choice or your child's choice to participate (or not) will not affect your child's grade in the course.

Additionally, as part of the TI study, your child's teacher will be videotaped during regular instruction, and it is possible that your child's likeness may be captured on video. Your child has the right to request that taping be stopped at any time. Video will be used for teacher data analysis only and will not be released or published. The results of the research study may be published at professional meetings and in research journals. To maintain confidentiality, your child will be assigned a code and his/her name will not be used. Records, data, and video will be stored in a locked cabinet in Padnos Hall at GVSU for 3 years after the close of the study and then destroyed. Furthermore, any student data used in publications or presentations will be anonymous or reported as class aggregate data.

The study has a possible benefit to your child since it will provide added practice taking standardized chemistry exams. The study has possible benefits to educators who design professional development programs for teachers, researchers who study teacher professional development, and high school chemistry teachers who use the materials generated by this project.

If you have any questions concerning the research study or your participation, please call us at (616) 331-3317.

Sincerely,

Deborah G. Herrington, Ph.D. Ellen J. Yezierski, Ph.D.
Target Inquiry Principal Investigator Target Inquiry Principal Investigator

I give consent for my child ___________________________ to participate in the above study.

_______________________________________________
Parent/Guardian Name

____________________________________      _________________________
Signature                      Date

If you have any questions about your rights as a research participant that have not been answered by the investigator, you may contact the Grand Valley State University Human Research Review Committee Chair as follows:
Paul J. Reitemeier, Ph.D., Chair, HRRC Office phone: (616) 331-3197 E-Mail: Reitemep@gvsu.edu

*Target Inquiry is funded by the National Science Foundation and the Camille and Henry Dreyfus Foundation

This research protocol has been approved by the Human Research Review Committee at Grand Valley State University.
PARENT PERMISSION LETTER PRE-MATERIALS STUDY

STUDENT ASSENT

I have been informed that my parent(s) has given permission for me to participate in a study that is investigating how a new teacher professional development program impacts student achievement. The study involves taking two tests. I understand that my teacher will be videotaped during regular instruction, and it is possible that my likeness be captured on video. I also understand that I have the right to request that taping be stopped at any time.

My participation in this project is voluntary and I have been told that I may stop my participation in this study at any time. If I choose not to participate, it will not affect my grade in any way.

________________________________________
Printed Name

________________________________________
Signature

________________________________________
School
Dear Parent,

Your child's chemistry teacher is a part of the Target Inquiry program at Grand Valley State University (GVSU). This program is designed to help teachers increase the quantity and quality of inquiry instruction in their chemistry classrooms. Research has shown that inquiry instruction can help students learn and retain chemistry concepts more effectively. We are conducting a study to determine how a new teacher professional development program affects teachers and student achievement in chemistry. Additionally, as part of this program, your child's teacher has developed new inquiry teaching materials that will be used in the classroom. These materials are aligned with the new Michigan High School Chemistry Content Expectations. To further improve his/her teaching, your child's teacher would like to collect data to evaluate the impact of these materials on students. These data may include student surveys, test results, or other classroom artifacts such as lab reports. The items that your child's teacher collects and the data analysis methods will depend on his/her student focus (e.g. motivation, conceptual understanding, data analysis skills, etc.).

We are requesting your child's participation. Your child's participation in this study is voluntary. You (or your child) are free to decide not to participate in this study or to withdraw at any time without adversely affecting your relationship with your teacher, the investigators, or GVSU. Your decision will not result in any loss of benefits to which your child is otherwise entitled. Specifically, your choice or your child's choice to participate (or not) will not affect your child's grade in the course.

Participation involves allowing the data from surveys, test results, or other classroom artifacts such as lab reports to be used in the analysis of the new classroom materials. Please note, that if you and your child choose for him/her not to participate, s/he is still responsible for completing the tests, assignments, or lab reports required for this course. However, his/her scores on such assignments will not be included in the data analysis. Additionally, as part of the TI study, your child's teacher will be videotaped during regular instruction, and it is possible that your child's likeness may be captured on video. Your child has the right to request that taping be stopped at any time. Video will be used for teacher data analysis only and will not be released or published. The results of the research study may be published at professional meetings and in research journals. To maintain confidentiality, your child will be assigned a code and his/her name will not be used. Records, data, and video will be stored in a locked cabinet in Padnos Hall at GVSU for 3 years after the close of the study and then destroyed. Furthermore, any student data used in publications or presentations will be anonymous or reported as class aggregate data.

The study has possible benefits to your child. First, it will provide them with added practice taking standardized chemistry exams. Second, your child will be engaged in learning activities that have been shown to improve student conceptual understanding and retention. The study also has possible benefits to educators who design professional development programs for teachers, researchers who study teacher professional development, and high school chemistry teachers who use the materials generated by this project.

If you have any questions concerning the research study or your participation, please call us at (616) 331-3317.

Sincerely,

Deborah G. Herrington, Ph.D.
Target Inquiry Principal Investigator

Ellen J. Yezierski, Ph.D.
Target Inquiry Principal Investigator

I give consent for my child ___________________________ to participate in the above study.

__________________________
Parent/Guardian Name

__________________________   __________________________
Signature                      Date

If you have any questions about your rights as a research participant that have not been answered by the investigator, you may contact the Grand Valley State University Human Research Review Committee Chair as follows:
Paul J. Reitemeier, Ph.D., Chair, HRRC Office phone: (616) 331-3197 E-Mail: Reitemep@gvsu.edu

*Target Inquiry is funded by the National Science Foundation and the Camille and Henry Dreyfus Foundation
This research protocol has been approved by the Human Research Review Committee at Grand Valley State University. File No. 25738 Expiration: 07/24/2010.

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STUDENT ASSENT

I have been informed that my parent(s) has given permission for me to participate in a study that is investigating how a new teacher professional development program impacts student achievement and how new inquiry teaching materials impact students. The study involves completing required surveys and/or course materials such as tests, assignments, or lab reports. I understand that my teacher will be videotaped during regular instruction, and it is possible that my likeness be captured on video. I also understand that I have the right to request that taping be stopped at any time.

My participation in this project is voluntary and I have been told that I may stop my participation in this study at any time. If I choose not to participate, it will not affect my grade in any way.

________________________________________
Printed Name

________________________________________
Signature

________________________________________
School
Dear [Name],

We are associate professors in the Department of Chemistry at Grand Valley State University (GVSU). We are conducting a research study to determine how Target Inquiry (a new professional development program) affects teachers, their teaching, and student achievement. This study will take place in your classroom and at GVSU (Padnos Hall), and you will be videotaped while teaching in your classroom for a maximum of 4 times during each school year you participate in the study.

We are requesting your participation, which will involve no more than 10 hours per year for two to five years beyond scheduled teaching time at your school and/or scheduled class time at GVSU. Your participation in this study is voluntary. You are free to decide not to participate in this study or to withdraw at any time without adversely affecting your relationship with the investigators or GVSU. Your decision will not result in any loss of benefits to which you are otherwise entitled. Specifically, your withdrawal from the study will not affect your grade or status in the chemistry concentration certificate program or M.Ed. program. In the case of early withdrawal stipends will be prorated. Additionally, you have the right to request that taping be stopped at any time.

The results of the research study will be published at professional meetings and in research journals. To maintain confidentiality, you will be assigned a code and your name will not be used. Video will be used for data analysis only and will not be released or published. Records, data, and video will be stored in a locked cabinet in Padnos Hall at GVSU for 3 years after the close of the study. After these data are analyzed by the researchers, tapes will be erased and records/data will be destroyed.

The study has many possible benefits to you since it involves your participation in a new professional development program. In the program, teachers can:

1) Expand their conceptual knowledge of chemistry by applying a research-oriented approach:
   a. Identify gaps in their knowledge and generate questions to address deficiencies.
   b. Access chemistry research literature and other appropriate resources.
   c. Collaborate with peers and scientists to refine and enhance understanding.

2) Develop a high school chemistry curriculum that is aligned with NSES and promotes the process of scientific inquiry.

3) Use action research to improve instruction and student learning by implementing and evaluating new teaching methods and curriculum.

You will also be exposed to research methods commonly used in quantitative and qualitative educational research. The study has possible benefits to those who design professional development programs for teachers, conduct research on professional development, and high school chemistry teachers who use the materials generated by this project.

If you have any questions concerning the research study or your participation, please call us at (616) 331-3317.

Sincerely,

Deborah G. Herrington, Ph.D.  Ellen J. Yezierski, Ph.D.
Target Inquiry Principal Investigator  Target Inquiry Principal Investigator

I give consent to participate in the study described above.

Name ____________________________  Signature ____________________________  Date ____________

This research protocol has been approved by the Human Research Review Committee at Grand Valley State University. File No. 25738 Expiration: 07/24/2010.
Dear

We are assistant professors in the Department of Chemistry at Grand Valley State University (GVSU). We are conducting a research study to determine how Target Inquiry (a new professional development program) affects teachers, their teaching, and student achievement. Your chemistry teacher, ____________________________, would like to participate. This study will take place in his/her classroom and at GVSU (Padnos Hall). S/he will be videotaped while teaching for a maximum of 4 times during each school year for a maximum of 5 years.

We are requesting your permission to conduct classroom observations and request student participation in the study. The GVSU Human Research Review Board has approved this study and the attached teacher, student assent, and parent permission forms. Attached is documentation of approval by the GVSU HRRC.

The results of the research study will be submitted for publication at professional meetings and in research journals. To maintain confidentiality, teachers and students will be assigned codes and their names will not be used. Video will be used for data analysis only and will not be released or published. Records, data, and video will be stored in a locked cabinet in Padnos Hall at GVSU for 3 years after the close of the study and then destroyed.

If you have any questions concerning the research study or your participation, please call us at (616) 331-3317.

Sincerely,

Deborah G. Herrington, Ph.D.  Ellen J. Yezierski, Ph.D.
Target Inquiry Principal Investigator  Target Inquiry Principal Investigator

I give consent to participate in the study described above.

________________________________________
School

________________________________________
Name

________________________________________
Signature  Date

If you have any questions that have not been answered by the investigator, you may contact the Grand Valley State University Human Research Review Committee Chair as follows:
Paul J. Reitemeier, Ph.D., Chair, HRRC  Office phone: (616) 331-3197  E-Mail: Reitemep@gvsu.edu

This research protocol has been approved by the Human Research Review Committee at Grand Valley State University. File No. 25738 Expiration: 07/24/2010.
Dear 

We are assistant professors in the Department of Chemistry at Grand Valley State University (GVSU). We are conducting a research study to determine how Target Inquiry (a new professional development program) affects teachers, their teaching, and student achievement. Your chemistry teacher, _____________, would like to participate. This study will take place in his/her classroom and at GVSU (Padnos Hall). S/he will be videotaped while teaching for a maximum of 4 times during each school year for a maximum of 5 years.

Additionally, as part of this program, ____ has developed new inquiry teaching materials that he/she will be implementing in his/her classroom. These materials are aligned with the new Michigan High School Chemistry Content Expectations. To further improve his/her teaching, ____ would like to collect data to evaluate the impact of these materials for students. This may include student surveys, test results, or other classroom artifacts such as lab reports.

We are requesting your permission to conduct classroom observations and request student participation in the study. The GVSU Human Research Review Board has approved this study and the attached teacher, student assent, and parent permission forms. Attached is documentation of approval by the GVSU HRRC.

The results of the research study will be submitted for publication at professional meetings and in research journals. ____’s new inquiry materials along with the results of their evaluation may be presented at conferences and/or published in educational journals such as the Science Teacher. To maintain confidentiality, teachers and students will be assigned codes and their names will not be used. Video will be used for data analysis only and will not be released or published. Records, data, and video will be stored in a locked cabinet in Padnos Hall at GVSU for 3 years after the close of the study and then destroyed. Furthermore, any student data used in the evaluation of the inquiry materials will be presented anonymously or as aggregate class data.

If you have any questions concerning the research study or your participation, please call us at (616) 331-3317.

Sincerely,

Deborah G. Herrington, Ph.D.                  Ellen J. Yezierski, Ph.D.
Target Inquiry Principal Investigator          Target Inquiry Principal Investigator

I give consent to participate in the study described above.

______________________________                     ______________________________
School                                                                                 Name

______________________________                     ______________________________
Signature                                                                               Date

If you have any questions that have not been answered by the investigator, you may contact the Grand Valley State University Human Research Review Committee Chair as follows:
Paul J. Reitemeier, Ph.D., Chair, HRRC       Office phone: (616) 331-3197      E-Mail: Reitemep@gvsu.edu

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